

# Chapter 5

- Born interpretation of the wave function  $\Psi$

$$P(x) dx = |\Psi(x,t)|^2 dx$$

is the probability that a particle will be found in an interval  $dx$  about the point  $x$ .

- Normalization Condition

$$\int_{-\infty}^{+\infty} |\Psi(x,t)|^2 dx = 1$$

and  $\int_a^b |\Psi(x,t)|^2 dx$  is the probability

of finding the particle in the interval

$$a \leq x \leq b.$$

• The wavefunction for a "free particle" is a "plane wave"

$$\psi(x,t) = e^{i(kx - \omega t)}$$

where  $k = \frac{p}{\hbar}$  ( $p$ : momentum of the particle)

and  $\omega = \frac{E}{\hbar}$  ( $E$ : kinetic energy of the particle)

⇒ This particle is **NOT LOCALIZED**

however its momentum ( $p = \hbar k$ ) and energy ( $E = \hbar \omega$ ) are **KNOWN PRECISELY**.

• If the particle is initially **LOCALIZED**, then the wavefunction must be a **WAVE PACKET**

$$\psi(x,t) = \int_{-\infty}^{+\infty} a(k) e^{i(kx - \omega t)} dk$$

↑  
amplitude

In the presence of a force Schrodinger equation is:  $(F = -\frac{dU}{dx})$

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x,t)}{\partial x^2} + U(x) \psi(x,t) = i\hbar \frac{\partial \psi(x,t)}{\partial t}$$

Stationary states are  $\psi(x,t) = \psi(x) e^{-i\omega t}$   
↑  
time independent  
Wave function

Schrodinger equation becomes

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + U(x) \psi(x) = E \psi(x)$$

← This is the time independent Schrodinger equation.

Note that  $|\psi(x,t)|^2 = |\psi(x)|^2 (e^{i\omega t} \cdot e^{-i\omega t})$   
 $= |\psi(x)|^2$

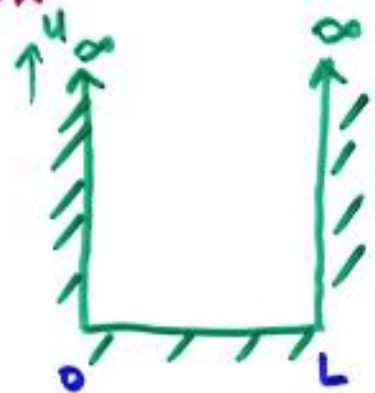
⇒ For stationary states the probabilities calculated from  $\psi(x,t)$  are time independent!



# Example #1: Particle in a box

$$U(x) = 0 \quad 0 \leq x \leq L$$

$$U(x) = \infty \quad \begin{array}{l} x < 0 \\ x > L \end{array}$$



- Outside the box;  $\psi(x) = 0$
- Inside the box;  $U(x) = 0$

$$\frac{d^2 \psi(x)}{dx^2} + \underbrace{\frac{2mE}{\hbar^2}}_{k^2} \psi(x) = 0$$

$$\boxed{\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)} \quad n = 1, 2, 3, \dots$$

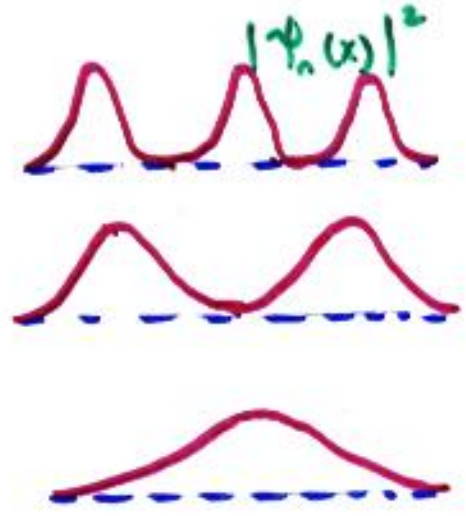
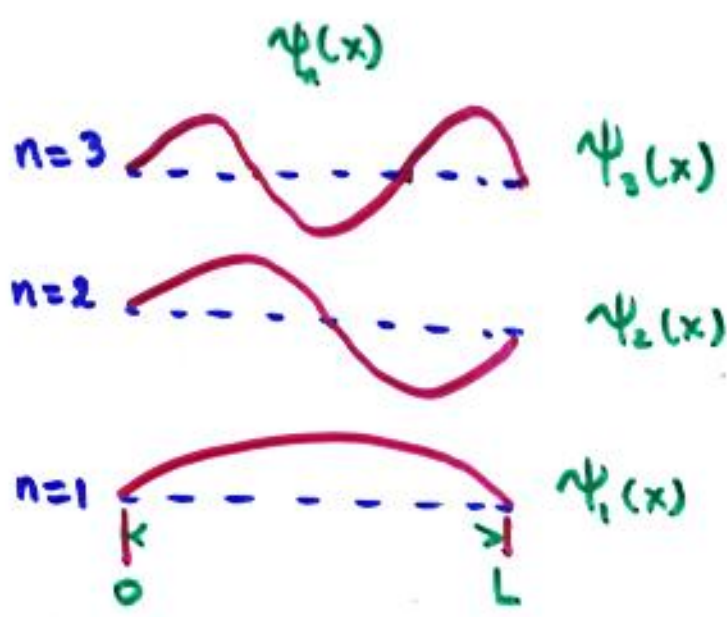
also:

$$\boxed{E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}} \quad n = 1, 2, 3, \dots$$

→ Energy is quantized !!!

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} \text{ is the } \underline{\text{ground state}}$$

$$E_n = n^2 E_1 \text{ are the } \underline{\text{excited states.}}$$



• For the harmonic oscillator  $U(x) = \frac{1}{2} m \omega^2 x^2$

Schrodinger equation is

$$\frac{d^2 \psi(x)}{dx^2} = \frac{2m}{\hbar^2} \left( \frac{1}{2} m \omega^2 x^2 - E \right) \psi(x)$$

The ground state wavefunction is

$$\psi_0(x) = \left( \frac{m\omega}{\pi\hbar} \right)^{1/4} e^{-\frac{m\omega}{2\hbar} x^2} = C_0 e^{-\alpha x^2}$$

The ground state energy is

$$E_0 = \frac{1}{2} \hbar \omega$$

The first excited state wavefunction is

$$\psi_1(x) = C_0 x e^{-\alpha x^2}$$

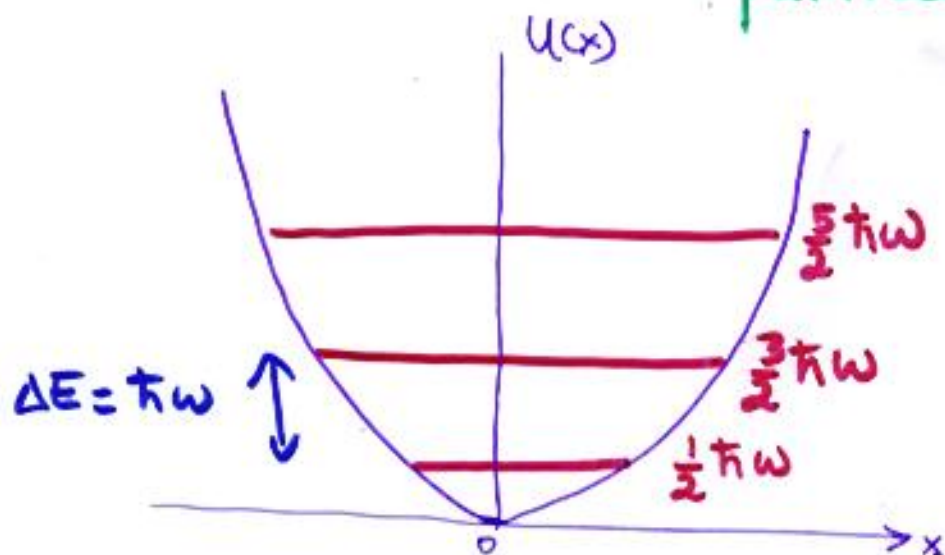
and

$$E_1 = \frac{3}{2} \hbar \omega$$

In general

$$E_n = \left(n + \frac{1}{2}\right) \hbar \omega \quad n=0, 1, 2, \dots$$

↑ Total energy of the particle.



Note that  $\Delta E = \hbar \omega = hf$  just like the energy of blackbody oscillators (Planck).



- The average position of a particle in quantum mechanics is called the expectation value  $\langle x \rangle$ .

$$\langle x \rangle = \int_{-\infty}^{+\infty} x |\psi|^2 dx$$

For any function  $f(x)$

$$\langle f \rangle = \int_{-\infty}^{+\infty} f(x) |\psi|^2 dx$$

If  $f(x) = x^2$

$$\langle x^2 \rangle = \int_{-\infty}^{+\infty} x^2 |\psi|^2 dx$$

The quantum uncertainty in particle position is

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

- In quantum mechanics there are two types of variables
  - sharp such as energy
  - fuzzy such as position

For sharp variables the quantum uncertainty = 0

An **observable** is any particle property that can be measured.

In Q.M. every observable  $\longrightarrow$  operator



$$\langle Q \rangle = \int_{-\infty}^{+\infty} \psi^* [Q] \psi dx$$

$\uparrow$  observable       $\uparrow$  operator

examples:

momentum  $[p] = -i\hbar \frac{\partial}{\partial x}$

$$[x^2] = x^2$$

Kinetic energy

$$[K] = \frac{[p]^2}{2m} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$

Hamiltonian

$$[H] = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + U(x)$$

energy

$$[E] = i\hbar \frac{\partial}{\partial t}$$