Comparison of CeBr₃ with LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl Detectors

Paul Guss^a, Michael Reed^a, Ding Yuan^a,

Matthew Cutler^b, Chris Contreras^b, and Denis Beller^b

^aRemote Sensing Laboratory, P.O. Box 98521, M/S RSL-13, Las Vegas, NV 89193,

gusspp@nv.doe.gov

bUniversity of Nevada, Las Vegas, 4505 S. Maryland Parkway, Las Vegas, NV 89154,

bellerd@unlv.nevada.edu

ABSTRACT

Energy resolution and detection efficiency were compared between two sizes of cerium bromide (CeBr₃) scintillators, three sizes of lanthanum bromide (LaBr₃:Ce) scintillators, three sizes of sodium iodide (NaI:Tl) scintillators, and a lanthanum chloride (LaCl₃:Ce) scintillator. Comparisons are made of key parameters such as energy resolution, detection efficiency, linearity, and self-activity of CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl scintillator detectors. The scintillator detectors are tested by comparing the peak separation and identification in the energy range up to 3.0 MeV using ¹³³Ba, ¹⁵²Eu, and naturally occurring radioactive materials [1]. The study has shown that CeBr₃ scintillator detectors provided by Saint-Gobain offer better resolution than NaI:Tl scintillator detectors. CeBr₃ detectors could resolve some closely spaced peaks from ¹³³Ba and ¹⁵²Eu, which NaI:Tl could not. LaBr₃:Ce has slightly better resolution, and a slightly higher efficiency than CeBr₃. In this work, "self-activity" of each of these four detector types was measured by operating the detectors themselves. A comparison of the intrinsic activity for all of the detectors in this study is demonstrated. For CeBr₃, the self-activity present may be reduced, or even eliminated in the future, through improved processes for growing the material. It will be discussed if, and under what conditions, CeBr₃ may be better than LaBr₃:Ce and LaCl₃:Ce for detection of certain special nuclear material γ-rays [2]. An overall advantage of CeBr₃ detectors over lanthanum halide and NaI:Tl detectors will be discussed.

Keywords: CeBr₃, LaBr₃:Ce, LaCl₃:Ce, NaI:Tl, efficiency, gamma spectral comparison, resolution.

1. INTRODUCTION

Comparisons of room-temperature cerium doped lanthanum bromide (LaBr₃:Ce), cerium doped lanthanum chloride (LaCl₃:Ce), and thallium doped sodium iodide (NaI:Tl) scintillator detectors have been made to cadminum zinc telluride (CZT) and high-purity germanium (HPGe) detectors [1]. In this work, these comparisons have been extended to include a relatively newer scintillator detector, cerium bromide (CeBr₃), based on interesting results obtained in recent studies [2]. New scintillators, activated with cerium (Ce) ions as the primary scintillation source such as LaBr₃:Ce and LaCl₃:Ce have gained interest due to good time resolution (100 to 600 picoseconds depending on crystal size), good energy resolution (2%-3% full width at half maximum [FWHM] at 661 keV), high efficiency and light output per keV (63 photons/keV), and fast decay time (16 ns) [1], [3]. These superior properties classify these materials as possible radiation detectors that could replace CZT and NaI:Tl for a variety of applications requiring detectors that operate at room temperature. The shortcoming of LaBr₃:Ce and LaCl₃:Ce scintillators is that they are slightly "self-active," which can be seen as spurious spectral lines when collecting a spectrum. These intrinsic spurious lines may limit the utilization of the detector, especially in low-level counting. The "self-activity" is due to the presence of the unstable 138La isotope, uranium, and actinium series contaminant [2], [4]-[5]. It has been mentioned that the uranium and actinium series contaminant in the most recent LaBr₃:Ce scintillators has been reduced greatly [4], [6]. In this work, "self-activity" of LaBr₃:Ce was measured by operating the detector itself. As it turns out, CeBr₃ shares many of the positive properties of LaBr₃:Ce and LaCl₃:Ce of good time resolution, good energy resolution, high efficiency and light output per keV, and fast decay time, but with a reduced self-activity [2]. Comparisons are made of key parameters such as energy resolution and detection efficiency of CeBr3, LaBr3; Ce, LaCl3; Ce, and NaI:Tl detectors. The scintillator detectors are tested by comparing the peak separation and identification in the energy range up to 1.5 MeV using ¹³³Ba, ¹⁵²Eu, and naturally occurring radioactive materials (NORM). The motivation of this study is to evaluate CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and

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2. EXPERIMENT

2.1 Scintillator Detectors

Part of this work included growing large crystals of the scintillator CeBr3 and evaluating their properties. Three CeBr3 crystals were grown and coupled to photomultiplier tubes; the first crystal measured $\frac{1}{2}$ " $\times \frac{1}{2}$ ", and the other two crystals were each 1" \times 1". Radiation Monitoring Devices, Inc. (RMD) has fabricated and used a ½" × ½" CeBr₃ crystal [10-11]. They provided their crystal to Saint-Gobain for this work in order to assist Saint-Gobain in growth and benchmarking of a new crystal for this study [12]. Fig. 1 is a picture of the ½" × ½" crystal grown at Saint-Gobain. All of these crystals were used for comparisons with LaBr3:Ce and NaI:T1 in each of three different size crystal dimensions. Another material, LaCl3:Ce, which included impurities of rare earth metals, was also compared for performance and self-activity [13]. In total, 10 different crystals are depicted in Table 1 for these experiments: $CeBr_3$ ($\frac{1}{2}$ " × $\frac{1}{2}$ "), $CeBr_3$ (1" × 1"), $LaBr_3$:Ce (1" × 1"), $LaBr_3$:Ce(2" × 2"), LaBr₃:Ce (3" × 3"), LaCl₃:Ce (1" × 1"), NaI:T1 (1" × 1"), NaI:T1 (2" \times 2"), and NaI:T1 (3" \times 3").



Figure 1. The $\frac{1}{2}$ " $\times \frac{1}{2}$ " CeBr₃ crystal epoxied to a photomultiplier tube [2].

Table 1. Summary of the different crystal sizes used in this study.

Crystal	½" ×½"	1" × 1"	2" × 2"	3"×3"	Total
CeBr ₃ 1 2	-3				
LaBr3:Ce	43	1			
LaCl3:Ce		1	1-		
NaI:T1	31	1 1			
Total	21	2	10		
Volume per Crystal (cc)	0.1	0.78	6.2	21.2	

2.2 Data Acquisiton

The scintillation detectors were coupled with a photomultiplier tube (for the $1" \times 1"$ cerium and lanthanum halide detectors a Photonis XP2060B photomultiplier tube was used) and a preamplifier was obtained from Saint-Gobain. The assembly was connected to and powered by the ORTEC DigiDART. Operation bias of the scintillators and the conversion gain were set to establish 1 keV per channel. A "self activity" spectrum was also taken of each of the scintillators, including NaI:Tl, for periods extending over one day in a low-background lead pyramid to illustrate the internal (self) detection of the intrinsic decay spectrum of La and possibly alpha particles from the uranium decay chain.

In this work, each of two sizes of CeBr₃ crystals were obtained and benchmarked against NaI:Tl and LaBr₃:Ce. This work also benchmarked a LaCl₃:Ce crystal. The team documented energy resolution and counting efficiency characteristics using a series of isotopes and γ -ray energies (22 Na, 60 Co, 137 Cs, 133 Ba, 152 Eu, 207 Bi, 228 Th, 241 Am, depleted uranium [DU], highly enriched uranium [HEU], and weapons-grade plutonium [Pu]). These isotopes gave a reasonable

coverage of energy range for overall performance evaluation. All sources used were "point sources." For each source spectrum, all radioisotopic sources were placed 40 cm from the detector inside a shielding safe with 1" of lead on all six faces of the safe. Both detector and source were located inside the lead safe. Source strength in the 1–100 μ Ci range was chosen in each case to highlight the difficulty of seeing the signal from a weak source in the presence of a high internal background. This simulates a situation as when a large amount of detector material (and hence large amounts of self-activity) is involved for detector configurations used for high sensitivity or long-range interrogation missions.

3. RESULTS AND DISCUSSION

3.1 Intrinsic Spectral Analysis

Preliminary estimates of self-activity, resolution, intrinsic efficiency, size effect, and linearity were defined [2].

Fig. 2 displays the background spectra obtained with the $1" \times 1"$ CeBr₃ detector taken inside the University of Nevada, Las Vegas (UNLV) Lead Shielding Pyramid, shown in Fig. 3, which was constructed for this project. It provided no less than 2" of lead on all sides of the detector. This count was taken for 115 hours in the low background UNLV Lead Shielding Pyramid (Fig. 3), and the spectrum in Fig. 2 represents the gross sum of counts for this period. Notice that the 40 K peak at 1461 keV stands out. Notice also the peaks at 1675 keV and 1903 keV. These peaks appeared in the spectrum for the $\frac{1}{2}" \times \frac{1}{2}"$ CeBr₃ detector as well. These two energy peaks do not appear to have an associated Compton edge, leading to speculation of some sort of correlation with α activity associated with the detector. There is not a clear or readily self-evident explanation for these peaks, making them an interesting area of study of the associated physics on its own merit.

Figs. 2 and 4 are spectra acquired for a 1" \times 1" CeBr $_3$ crystal and a 1" \times 1" LaBr $_3$:Ce crystal, respectively, for comparison. Fig. 4 shows the background spectrum measured with the LaBr $_3$:Ce detector after over 67 hours in the low background UNLV Lead Shielding Pyramid. Three intrinsically produced photon peaks from the decay of ¹³⁸La were observed. These are the 789 keV γ line from the β decay branch, the 1436 keV γ line from the electron capture branch, and a 32 keV X-ray fluorescence peak. A 1461 keV line from ⁴⁰K, with a much smaller scale, is present (i.e., see the measurement made in Fig. 2 without the ¹³⁸La background lines) but difficult to observe next to the 1436 keV line. One immediately sees the features associated with the intrinsic activity for LaBr $_3$:Ce from the background count. These features include the 789 keV γ -ray peak, along with an associated β decay spectrum (0–255 keV), and the 1436 keV γ -ray peaks, along with an associated a 32 keV X-ray. In fact, the 32 keV X-ray may be emitted nearly in coincidence with 1436 keV γ -ray, creating a comingled signature at the sum energy of 1468 keV in proximity to the 1436 keV line and the ⁴⁰K line at 1461 keV, leading to the rather large width of this "peak." The same phenomena of near coincidence signatures occurs between the 789 keV γ -ray and the β decay spectrum (0–255 keV), leading to the feature between 789 keV and 1 MeV, which is hardy Gaussian in shape yet is predominately correlated to a monoenergetic 789 keV γ -ray.

The "self activity" and background spectrum of the LaBr₃:Ce scintillator is shown in Fig. 4. The continuum with an endpoint at approximately 255 keV is probably due to the β decay of ¹³⁸La to ¹³⁸Ce. The emission line with a peak around 1461 keV corresponds to the overlapped 1461 keV line of ⁴⁰K, originating from the glass of the photomultiplier tube, with the 1436 keV ¹³⁸La line summing internally with 32 keV X-rays. The 789 keV line sitting on the β continuum is only just resolvable. The low-energy photon peak at approximately 32 keV is due to ¹³⁸Ba K_{α 1} (32.2 keV) and K_{α 2} (31.8 keV) X-ray fluorescence. ¹³⁸Ba is the daughter product of ¹³⁸La from the electron capture branch. The broad band peaks (between 1700 and 2700 keV) due to alpha particles from the uranium decay chain [2], [4]–[5] are also observed (Fig. 2).

In Fig. 5, results for the self-activity are shown as a function of the size of the LaBr₃:Ce crystal. Fig. 6 is a compelling spectral comparison that argues for CeBr₃. Our work documented a lower background with CeBr₃ than with either LaBr₃:Ce or LaCl₃:Ce, for the energy region of interest when looking for γ -rays associated with manmade radioisotopes. The intrinsic background observed in the CeBr₃ spectra is mostly due to contaminants associated with α signatures that are evident in Fig. 2. In the critical low-energy areas of the spectrum, where the lower energy special nuclear material (SNM) peaks occur, the CeBr₃ had a reduction in intrinsic background.

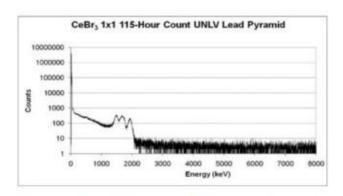


Figure 2. Background gamma-ray spectrum using the 1" × 1" CeBr₃ detector taken inside the UNLV Lead Shielding Pyramid. This count was taken for 115 hours.



Figure 3. Background and self-activity gamma-ray spectra are acquired at UNLV using this "UNLV Lead Shielding Pyramid.

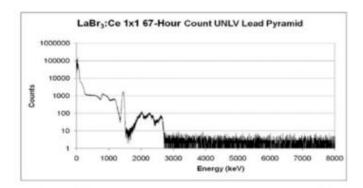


Figure 4. Background gamma-ray spectrum using the 1" × 1" LaBr₃:Ce detector taken inside the UNLV Lead Shielding Pyramid. This count was taken for 67 hours Pyramid."

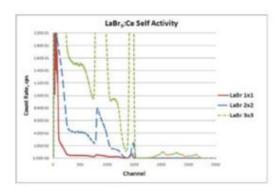


Figure 5. Background and self-activity gammaray spectra acquired for 3 sizes of LaBr₃:Ce detectors using the "UNLV Lead Shielding

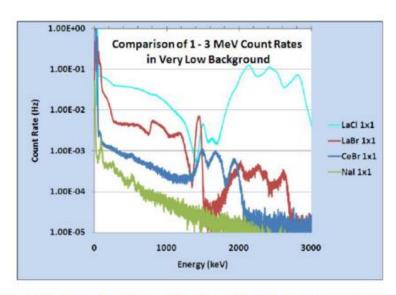


Figure 6. Comparison of the intrinsic "self-activity" for each of the CeBr3, LaBr3:Ce, LaCl3:Ce, and NaI:Tl scintillator detectors [2].

3.2 Energy Resolution Comparison

The energy resolutions of CeBr₃, LaBr₃:Ce, LaCl₃:Ce were compared with the resolution of NaI:Tl. Before discussing energy resolution, we would like to note that generally smaller scintillation crystals produce a better resolution. Also good spectral resolution is required for isotopic identification, particularly when peaks are closely grouped together as in HEU and Pu, or for identification of isotopes in mixed sources.

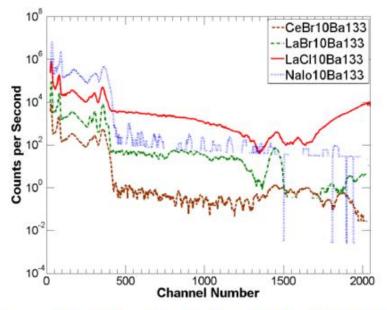


Figure 7. 133 Ba (6 μ Ci) energy spectrum comparison between 1" \times 1" CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl detectors [2]. Each detector is plotted on a successive decade to allow comparison. To obtain actual number of counts measured per second, divide CeBr₃ spectrum by 10^3 , LaBr₃:Ce spectrum by 10^4 , LaCl₃:Ce spectrum by 10^5 , and NaI:Tl spectrum by 10^6 .

For example, 133 Ba has four main peaks grouped closely together (separated by less than 100 keV) showing the importance of adequate spectral resolution for isotopic identification. In Fig. 7, the NaI:T1 detector is unable to resolve all four key peaks of 133 Ba, namely 276, 303, 356, and 383 keV. The LaBr₃:Ce and LaCl₃:Ce detectors are able to resolve these emission line groupings. The four peaks could be extracted from the CeBr₃ if good peak fitting routines are used. Interestingly in Fig. 7, it can be observed that when the source is sufficiently small enough (6 μ Ci of 133 Ba), then the curves of the NaI:T1 and LaBr₃:Ce cross, even though the NaI:T1 is plotted against a higher decade. This is indicative of the high values of "self-activity" of the lanthanum halide detectors and of how this self-activity may reduce sensitivity to very small sources.

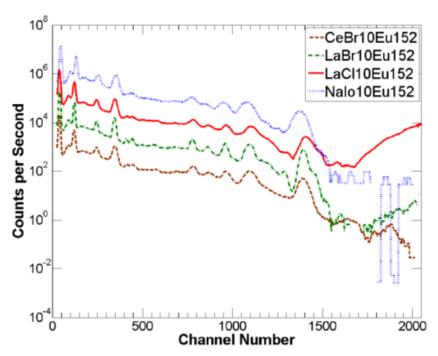


Figure 8. ¹⁵²Eu (24 µCi) energy spectrum comparison between 1" × 1" CeBr₃, LaBr₃:Ce, LaBr₃:Ce, and NaI:Tl detectors [2]. Each detector is plotted on a successive decade to allow comparison. To obtain actual number of counts measured per second, divide CeBr₃ spectrum by 10³, LaBr₃:Ce spectrum by 10⁴, LaCl₃:Ce spectrum by 10⁵, and NaI:Tl spectrum by 10⁶.

Good spectral resolution and high efficiency becomes important when working with sources that have a complicated decay scheme. For example, 152 Eu has a complicated decay scheme that has emissions over a large range of energy, with relatively close peaks at both ends of this energy range. As seen in Fig. 8, CeBr₃, LaBr₃:Ce, and LaCl₃:Ce were able to resolve closely spaced emission lines of Eu, such as 411 and 444 keV lines. When a similar 152 Eu spectrum was taken with NaI:Tl, these close spectral lines could not be resolved. Notice also that the 1436 keV related cluster associated with the lanthanum halide detectors interferes with production of a clean Gaussian peak for the 1408 keV γ -ray line in 152 Eu spectra, whereas this peak is more clearly defined without interference using the CeBr₃ or the NaI:Tl detectors. In Figs. 9 and 10, comparisons are made for DU and HEU, respectively. Fig. 11 shows the greater ability of the CeBr₃, LaBr₃:Ce, and LaCl₃:Ce detectors over the NaI:Tl detector in resolving the 239 Pu γ -ray photopeaks in the region of 300 to 414 keV. For the 239 Pu isotope, the peaks were on a larger background when acquired with the LaBr₃:Ce or a LaCl₃:Ce crystal. The signal-to-noise ratio was better for the 1" \times 1" CeBr₃ detector than for these other two detectors. Fig. 12 illustrates the variations of energy resolution (defined as FWHM/E) of the detectors as a function of the γ energies. The best fit of the CeBr₃, LaBr₃:Ce, and LaCl₃:Ce data is shown in Fig. 12, providing another comparison between the CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl detectors. Our fit is not an ideal inversely proportional function of the square root of the energy [5, 9].

In general, CeBr₃ is a good compromise between the LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl detectors with CeBr₃, LaBr₃:Ce, and LaCl₃:Ce slightly outperforming NaI:Tl. As seen in Fig. 12, the resolution of the CeBr₃, LaBr₃:Ce, and LaCl₃:Ce detectors appears to be excellent for energies above approximately 120 keV; whereas below 120 keV, the resolution becomes similar to NaI:Tl, as described in the publication [5]. For energies below 120 keV, CeBr₃, LaBr₃:Ce, and LaCl₃:Ce offer little advantage over NaI:Tl. Moreover, when considering the cost of the scintillator, NaI:Tl could be preferable over CeBr₃, LaBr₃:Ce, and LaCl₃:Ce. The resolution of the ¹³⁷Cs 662 keV line for NaI:Tl is 7% compared with 3.4% for LaBr₃:Ce, 4.4% for LaCl₃:Ce, and 5.8% for CeBr₃ (Fig. 12). This demonstrates that CeBr₃, LaBr₃:Ce, and LaCl₃:Ce detectors have clear advantages over the properties of the NaI:Tl detector. Others have reported measuring the ¹³⁷Cs 662 keV line with a 4.9% resolution using CeBr₃ [14]. With regard to naturally occurring radioactive materials, a leading concern with the lanthanum halides is the interference between the cluster of 1436 keV to 1468 keV signatures due to ¹³⁸La and ⁴⁰K.

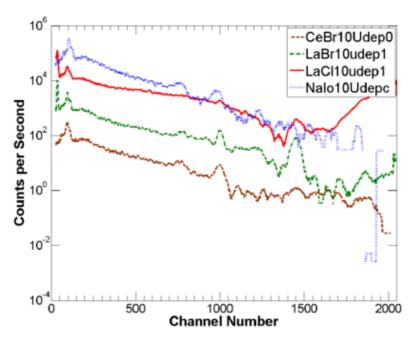


Figure 9. DU (154 μ Ci) energy spectrum comparison between 1" × 1" CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl detectors [2]. Each detector is plotted on a successive decade to allow comparison. To obtain actual number of counts measured per second, divide CeBr₃ spectrum by 10³, LaBr₃:Ce spectrum by 10⁴, LaCl₃:Ce spectrum by 10⁵, and NaI:Tl spectrum by 10⁶.

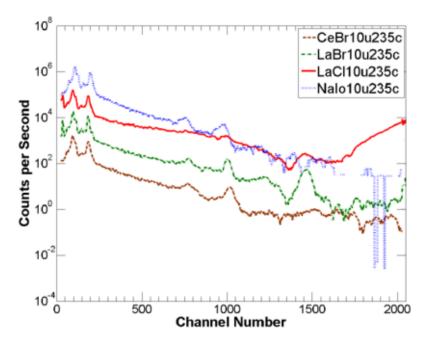


Figure 10. 235 U (100 μ Ci) energy spectrum comparison between 1" \times 1" CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl detectors [2]. Each detector is plotted on a successive decade to allow comparison. To obtain actual number of counts measured per second, divide CeBr₃ spectrum by 10³, LaBr₃:Ce spectrum by 10⁴, LaCl₃:Ce spectrum by 10⁵, and NaI:Tl spectrum by 10⁶.

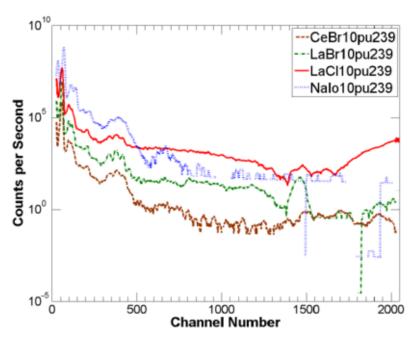


Figure 11. 239 Pu (2 μ Ci) energy spectrum comparison between 1" \times 1" CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:T1 detectors [2]. Each detector is plotted on a successive decade to allow comparison. To obtain actual number of counts measured per second, divide CeBr₃ spectrum by 10^3 , LaBr₃:Ce spectrum by 10^4 , LaCl₃:Ce spectrum by 10^5 , and NaI:T1 spectrum by 10^6 .

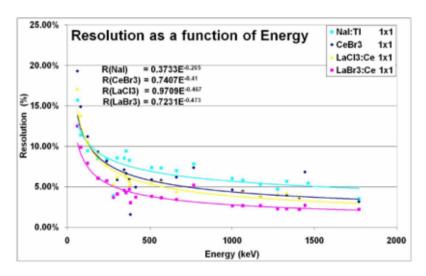


Figure 12. Energy resolution comparison between 1" × 1" CeBr3, LaBr3:Ce, LaCl3:Ce, and NaI:T1 detectors [2].

3.3 Detection Efficiency Comparison

A comparison of the full energy peak relative intrinsic efficiency is shown in Fig. 13 for the four 1" × 1" CeBr₃, LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl scintillator detectors in the energy range of 122 to 1408 keV. Figs. 14 and 15 plot the intrinsic efficiencies. Note that the larger detectors have a greater efficiency because their volume is larger. When comparing LaBr₃:Ce with LaCl₃:Ce, LaBr₃:Ce has a greater efficiency. When examining the data for CeBr₃, note that its efficiency is comparable to LaBr₃:Ce and better than the other two scintillators. In the high-energy regime (above approximately 1 MeV), the efficiency of the scintillators are similar.

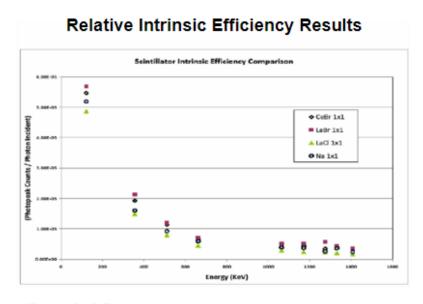


Figure 13. Relative intrinsic efficiency for CeBr3 is comparable to other detector materials in class.

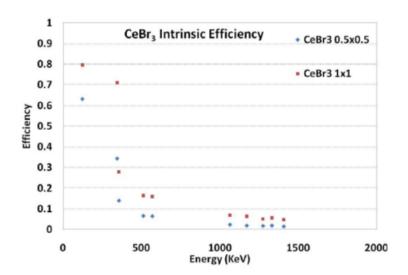


Figure 14. Intrinsic efficiency for CeBr3 does increase with size of the crystal.

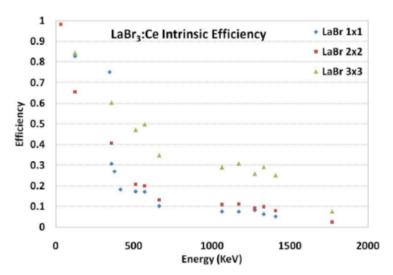


Figure 15. Intrinsic efficiency for LaBr3: Ce also increases with size of the crystal.

3.4 Linearity

We measured the non-linearity of photon yield versus energy for the CeBr₃ detectors and determined the linearity (Fig.16) to be better than 2% in the 100-1500 keV energy range. The NaI:Tl non-linearity is an order of magnitude worse than for CeBr₃. This work also showed that the non-linearity for LaBr₃:Ce was somewhere in between these two extremes. This finding for LaBr₃:Ce is consistent with what has been reported [15]. We prepared a few additional plots as a linearity analysis in Figs. 17-23 for the $\frac{1}{2}$ " \times $\frac{1}{2}$ " CeBr₃, the 1" \times 1" NaI:Tl, and the 3" \times 3" LaBr₃:Ce detectors and will give a brief interpretation of them.

Fig. 16, shows a plot of photon energy E_{γ} vs. Centroid channel number $Ch_{centroid}$ for the $\frac{1}{2}$ " $\times \frac{1}{2}$ " $CeBr_3$ and 1" $\times 1$ " NaI:Tl detectors. Here the $Ch_{centroid}$ number is the channel number for which the centroid of the peak corresponding to E_{γ}

appears. We were investigating linearity (light yield) here. The problem with the plots that we made is that one cannot tell much by looking at them. To this end, we plotted the following:

Where, E_{γ} is the known photon energy (γ or x-ray) arising from the measured source, and the value of the parameter \mathcal{F} is defined as:

$$\mathcal{F} = (E_{\gamma} - Ch_{centroid}) / E_{\gamma}$$

This is a normalized residual of sorts. It is more meaningful if one looks at the plot (Fig. 17) of this figure of merit. Notice that the signs change and recall that, with a calibration of 1 keV per channel, the photopeak energy for the 662 keV ¹³⁷Cs γ has been placed in Channel 662. This means that the quasi-residual is 0 for ¹³⁷Cs. In instances where the value of F is negative, the peak centroid appeared to the right of where it would be with a linear detector. This means that there was proportionally more light than what would have been expected based on the 137Cs light yield. In instances where the value of F is positive, the peak centroid appeared to the left of where it would be with a perfectly linear detector. This of course means that there was proportionally less light than what would have been expected based on the 137Cs light yield. This method is a much better way of exhibiting linearity. It is normalized because, while being 5 channels off at ²⁴¹Am (60 keV) is unacceptable, at the same time 5 channels off at ²²⁸Th (2.614 MeV) implies a smaller relative error in energy. Something to note about this is that for the plots in which there occurs many sign changes in the value of F, one can conclude that the emission spectra itself is erratic. For instances in which the plot is very smooth (the NaI:T1 plot shown in Fig. 19), one can conclude that the light yield curve, although not extremely linear, is quite smooth. Specifically in regard to detector comparisons, Fig. 18 shows an absolute variance from linearity of up to ±5 channels for CeBr₃ over the 0-2 MeV energy range. However, a variance of over 40 channels is evident in Fig. 20 for the case of NaI:T1 over the same energy range, with Fig. 22 showing that LaBr3:Ce is somewhere in between these extremes. The scatter in the non-linearity residues is much larger for NaI:Tl than for CeBr3 These data indicate that the CeBr₃ detector is the most linear of those studied in this work over the 0-2 MeV energy range (using the experimental setup conditions described above, which results in a 1 keV/channel calibration).

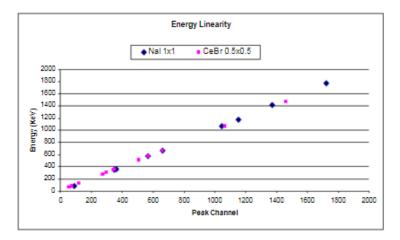


Figure 16. Linearity Plots of the NaI:Tl 1" \times 1" and CeBr₃ $\frac{1}{2}$ " \times $\frac{1}{2}$ " detector. Here the peak channel is plotted vs. the energy of the photon.

Normalized Linearity Analysis of CeBr₃ 0.5" x 0.5"

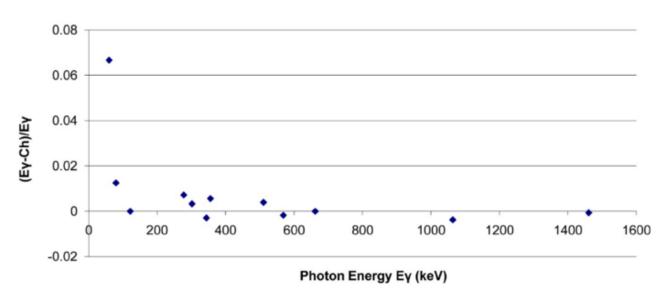


Figure 17. Normalized Linearity Analysis of the $\frac{1}{2}$ " CeBr₃ detector. Here the difference of the photon energy and channel divided by the photon energy is plotted against the photon energy.

Linearity Analysis of CeBr₃ 0.5" x 0.5"

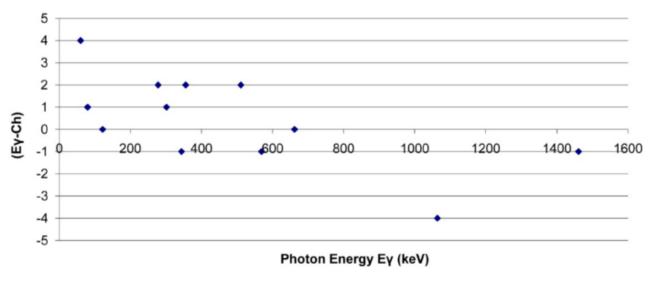


Figure 18. Linearity Analysis of the $\frac{1}{2}$ " \times $\frac{1}{2}$ " CeBr₃ detector. Here the difference of the photon energy and channel is plotted against the photon energy.

Normalized Linearity Analysis of 1" x 1" Nal

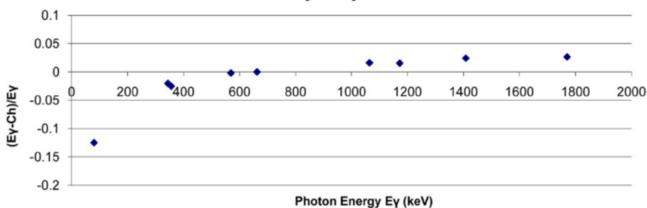


Figure 19. Normalized Linearity Analysis of the $1'' \times 1''$ Nal detector. Here the difference of the photon energy and channel divided by the photon energy is plotted against the photon energy.

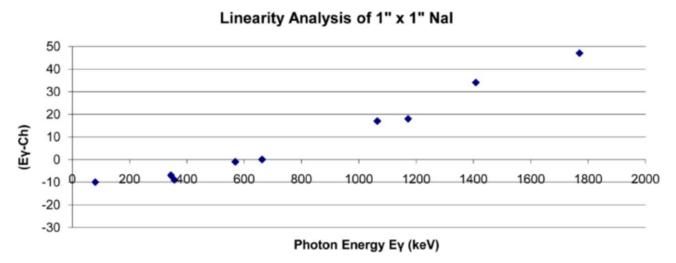


Figure 20. Linearity Analysis of the $1'' \times 1''$ NaI:T1 detector. Here the difference of the photon energy and channel is plotted against the photon energy.

Normalized Linearity Analysis of 3" x 3" LaBr₃

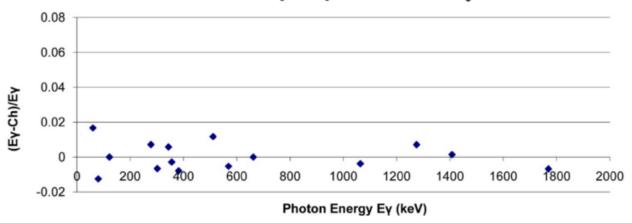


Figure 21. Normalized Linearity Analysis of the $3'' \times 3''$ LaBr₃:Ce detector. Here the difference of the photon energy and channel divided by the photon energy is plotted against the photon energy.

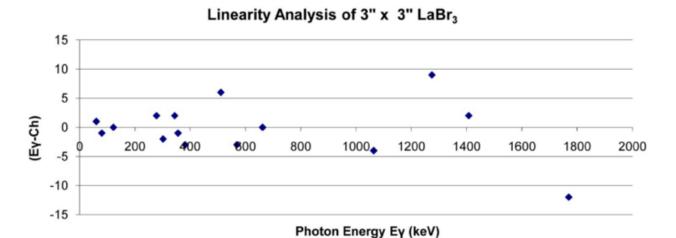


Figure 22. Linearity Analysis of the $3'' \times 3''$ LaBr₃:Ce detector. Here the difference of the photon energy and channel is plotted against the photon energy.

4. SUMMARY

This work started out as an investigation of room-temperature higher resolution detectors for use in extremely high sensitivity or signal low count rate nonproliferation missions. To that end, the results indicated that the CeBr₃ detector is as good as or better than the LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl scintillator detectors when looking for very low count rate signatures of low energy SNM γ -rays. CeBr₃ appears to be an excellent alternative for LaBr₃:Ce, LaCl₃:Ce, and NaI:Tl scintillator detectors. It could also be a possible replacement for CZT Te in a variety of applications for γ -ray spectroscopy, in particular where moderate energy resolution and high detection efficiency is required. CeBr₃ has good linearity of response when compared to these other detectors. Also evident for the lanthanum halide detectors, for long counting times, are three intrinsic characteristic spectral lines (2 γ and 1 X-ray) and two 255 keV endpoint β spectra (one spectrum from 0 to 255 keV and the other from 789 to 1044 keV). These are associated with ¹³⁸La. However, being aware of these emission lines and β spectra, γ -ray spectroscopy using LaBr₃:Ce and LaCl₃:Ce should enable the experimenter to correct for many applications. However, for the most sensitive of measurements, best results with background in the energy regions associated with these ¹³⁸La-related self-activities can be obtained easily by replacing the lanthanum halide detector with CeBr₃. CeBr₃ scintillators can be used in a wide range of applications, such as coincidence counting, isotopic identification, and safeguards applications for monitoring of nuclear materials.

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