

CH 11 The Energy Equation

①

Consider a stationary control volume through which a fluid is moving. The general energy balance equation :

$$\left\{ \begin{array}{l} \text{Rate of} \\ \text{Accumulation.} \\ \underline{KE + Int. E} \end{array} \right\} = \left\{ \begin{array}{l} \text{Rate of Energy} \\ \text{Addition by} \\ \text{Convection.} \\ \underline{KE. + Int. E} \end{array} \right\} + \left\{ \begin{array}{l} \text{Rate of Energy} \\ \text{Addition by} \\ \text{Molecular Transf.} \\ \text{Conduction} \end{array} \right\} + \\
 \left\{ \begin{array}{l} \text{Rate of Work} \\ \text{done on system} \\ \text{by molecular} \\ \text{Mechanisms} \\ (\text{i.e. stress + pressure}) \end{array} \right\} + \left\{ \begin{array}{l} \text{Rate of work} \\ \text{done on system} \\ \text{by external} \\ \text{forces (gravity)} \end{array} \right\}$$

before we proceed we introduce the Combined Energy Flux Vector \vec{e} :

$$\vec{e} \left(\frac{\text{J}}{\text{m}^2 \text{s}} \right) = \frac{1}{2} \rho v^2 \vec{v} + \rho \vec{U} \vec{v} + p \vec{v} + \underline{\underline{\epsilon}} \cdot \vec{v} + \vec{q}$$

↓ KE ↓ Int. E. ↓ work due to pressure ↓ work due to stress ↓ conduction

note : $v^2 = \vec{v} \cdot \vec{v} = v_x^2 + v_y^2 + v_z^2$

add all terms :

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$$\Delta x \Delta y \Delta z \frac{\partial}{\partial t} \left(\frac{1}{2} \sigma v^2 + \sigma \hat{U} \right) = \left(e_x |_x - e_x |_{x+\Delta x} \right) \Delta y \Delta z + \left(k_y |_y - k_y |_{y+\Delta y} \right) \Delta x \Delta z \\ + \left(e_z |_z - e_z |_{z+\Delta z} \right) \Delta x \Delta y \\ + \sigma \Delta x \Delta y \Delta z (\vec{V} \cdot \vec{g})$$

divide by $\Delta x \Delta y \Delta z$ and $\Delta x \Delta y \Delta z \rightarrow 0$

$$\frac{\partial}{\partial t} \left(\frac{1}{2} \sigma v^2 + \sigma \hat{U} \right) = - \vec{\nabla} \cdot \vec{e} + \sigma (\vec{g} \cdot \vec{V}) \left(\frac{J}{m^3 s} \right)$$

recall definition of \vec{e} (energy flux vector)

$$\frac{\partial}{\partial t} \left(\frac{1}{2} \sigma v^2 + \sigma \hat{U} \right) = - \vec{\nabla} \cdot \left(\frac{1}{2} \sigma v^2 + \sigma \hat{U} \right) \vec{V} \quad \text{convection} \\ - \vec{\nabla} \cdot \vec{q} \quad \text{conduction} \\ - \vec{\nabla} \cdot \rho \vec{V} \quad \text{Work due to pressure} \\ - \vec{\nabla} \cdot [\underline{\underline{c}} \cdot \vec{V}] \quad \text{Work due to friction} \\ + \sigma (\vec{V} \cdot \vec{g}) \quad \text{Work by external forces}$$

... . Eq. (1)

Recall the equation of motion :

$$\frac{\partial}{\partial t} \rho \vec{v} = - \vec{\nabla} \cdot \rho \vec{v} - \vec{\nabla} p - \vec{\nabla} \cdot \underline{\underline{\epsilon}} + \rho g \quad \left(\frac{N}{m^3} \right)$$

take dot product with \vec{v}

0
0
0

$$\boxed{\frac{\partial}{\partial t} \left(\frac{1}{2} \rho v^2 \right) = - \vec{\nabla} \cdot \frac{1}{2} \rho v^2 \vec{v} - \vec{\nabla} \cdot p \vec{v} - p (- \vec{\nabla} \cdot \vec{v})} \quad \text{Eq. ②}$$

$$- \vec{\nabla} \cdot (\underline{\underline{\epsilon}} \cdot \vec{v}) - (- \vec{\epsilon} : \vec{\nabla} \vec{v}) + \rho (\vec{v} \cdot \vec{g})$$

$\left(\frac{J}{m^3 s} \right)$

Eq. ① Combines both mechanical energy and internal energy

Eq. ② Mechanical Energy equation

Eq. ① - Eq. ② :

$$\boxed{\frac{\partial}{\partial t} (\rho \hat{U}) = - \vec{\nabla} \cdot \rho \hat{U} \vec{v} - \vec{\nabla} \cdot q - p (\vec{\nabla} \cdot \vec{v}) - \underline{\underline{\epsilon}} : \vec{\nabla} \vec{v}}$$

..... Eq. ③

Equation ③ Internal Energy equation and it is useful to obtain energy equation in terms of temperature

Rearranging by utilizing substantial derivative ⁽⁴⁾

$$\rho \frac{D\hat{U}}{Dt} = - \vec{\nabla} \cdot \vec{q} - P(\vec{\nabla} \cdot \vec{v}) - \underline{\underline{\epsilon}} : \underline{\underline{\nabla}} v$$

Enthalpy $\hat{H} = \hat{U} + P\hat{V}$
 specific volume ($\frac{1}{\rho}$)

$$\rho \frac{D\hat{H}}{Dt} = - (\vec{\nabla} \cdot \vec{q}) - (\underline{\underline{\epsilon}} : \underline{\underline{\nabla}} v) + \frac{DP}{Dt}$$

recall from thermodynamics :

$$\begin{aligned} d\hat{H} &= C_p dT + \left[\hat{V} - T \left(\frac{\partial \hat{V}}{\partial T} \right)_P \right] dp \\ &= C_p dT + \left[\frac{1}{\rho} + \left(\frac{\partial \ln \rho}{\partial \ln T} \right)_P \right] dp \end{aligned}$$

$\gamma = 1$ for ideal gas.

$$\Rightarrow \rho C_p \frac{DT}{Dt} = - \vec{\nabla} \cdot \vec{q} - (\underline{\underline{\epsilon}} : \underline{\underline{\nabla}} v) - \left(\left(\frac{\partial \ln \rho}{\partial \ln T} \right)_P \frac{DP}{Dt} \right)$$

The Energy Equation.

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Special Forms of Energy Equation:

① Constant density fluids

$$\cancel{\rho} \cancel{C_p} \frac{DT}{Dt} = -\nabla \cdot \vec{q} - \cancel{\dot{\Sigma}} : \nabla \underline{v}$$

② Newtonian Fluids:

$$\cancel{\dot{\Sigma}} : \nabla \underline{v} = \mu \vec{\Phi}_v + \kappa \Psi \quad , \quad \Psi = (\nabla \cdot \underline{v})^2$$

see Appendix · B · 7

$$\cancel{\rho} \cancel{C_p} \frac{DT}{Dt} = -\nabla \cdot \vec{q} - (\mu \vec{\Phi}_v + \kappa \Psi) - \left(\frac{\delta \ln \Psi}{\delta \ln T} \right)_P \frac{DP}{Dt}$$

③ Materials following Fourier's law of conduction

$$\vec{q} = -\kappa \vec{\nabla} T$$

if κ is const

$$\cancel{\rho} \cancel{C_p} \frac{DT}{Dt} = \kappa \nabla^2 T - (\mu \vec{\Phi}_v + \kappa \Psi) - \left(\frac{\delta \ln \Psi}{\delta \ln T} \right)_P \frac{DP}{Dt}$$

④ Const. P , Newtonian, Fourier's law & Incompressible flow

$$\cancel{\rho} \cancel{C_p} \frac{DT}{Dt} = \kappa \nabla^2 T - \mu \vec{\Phi}_v$$
