

Q) Use the Hirschfelder et al. equation to 1/2
 calculate diffusion coefficient of CO_2 in N_2 at
 298 K ϵ_1 100 kPa.

Solution :-

$$\text{eq. 24.33} \quad D_{AB} = \frac{0.001858 T^{3/2} \left[\frac{1}{M_A} + \frac{1}{M_B} \right]^{1/2}}{P \delta_{AB}^2 S_D}$$

$$T = 298 \text{ K} \quad , \quad P = 100 \text{ kPa} = 1 \text{ atm}$$

$$M_A = M_{\text{CO}_2} = 44 \quad , \quad M_B = M_{\text{N}_2} = 28$$

$$\text{eq. 24.40} \quad \epsilon_{AB} = \sqrt{\epsilon_A \epsilon_B}$$

Appendix K 2

$$\delta_A = \delta_{\text{CO}_2} = 3.996^\circ \text{A} \quad , \quad \delta_B = \delta_{\text{N}_2} = 3.681^\circ \text{A}$$

$$\delta_{AB} = \frac{3.681 + 3.996}{2} = 3.8385$$

$$\frac{\epsilon_B}{K} = 91.5 \quad , \quad \frac{\epsilon_A}{K} = 190 \quad (\text{CO}_2)$$

(N₂)

$$\text{where } K = 1.38 \times 10^{-16} \text{ ergs/K}$$

$$\epsilon_{AB} = \sqrt{\frac{91.5 \times 190}{K^2}} = \frac{131.85}{K}$$

$$\therefore \frac{KT}{\epsilon} = \frac{298}{131.85} = 2.26$$

From Appendix K1 at $kT/\epsilon = 2.26$

(2/2)

$$S_D = \frac{1.026 + 1.012}{2} = 1.019$$

\therefore

$$D_{AB} = 0.001858 (298)^{3/2} \left[\frac{1}{44} + \frac{1}{28} \right]^{1/2}$$

$$(1) (3.8385)^2 (1.019)$$

$$D_{AB} = 0.154 \text{ cm}^2/\text{s}$$