

CH #39

h.w. Solution

#2

$$E = \left(\frac{h^2}{8mL^2} \right) n^2$$

$$E_{\text{ground}} = \left(\frac{h^2}{8mL^2} \right)$$

a.

$$L = 2 \times 10^{-10} \text{ m}$$

$$m_e = 9.1 \times 10^{-31} \text{ kg}$$

b.

$$m_p = 1.67 \times 10^{-27} \text{ kg.}$$

$$\#9. \quad E_n = \left(\frac{h^2}{8mL^2} \right) n^2$$

$$a. \quad E_n = hf = \frac{hc}{\lambda}$$

$$\lambda = \frac{hc}{E_n}$$

$$a. \quad \lambda_1 = \frac{hc}{E_1}$$

$$b. \quad \lambda_2 = \frac{hc}{E_2}$$

$$c. \quad \lambda_3 = \frac{hc}{E_3}$$

#13.

Probability of finding the electron between x_1 and x_2

$$= \int_{x_1}^{x_2} A^2 \sin^2\left(\frac{n\pi}{L}x\right) dx$$

$$= A^2 \int_{x_1}^{x_2} \sin^2(ax) dx, \quad a = \frac{n\pi}{L}$$

$$A = \sqrt{\frac{2}{L}}$$

$$= A^2 \left[\int \right]$$

$$\cos 2ax = 1 - 2\sin^2 ax$$

$$\sin^2 ax = \frac{(1 - \cos 2ax)}{2}$$

$$\therefore A^2 \int_{x_1}^{x_2} \sin^2 ax dx = \frac{A^2}{2} \int (1 - \cos 2ax) dx$$

$$= \frac{A^2}{2} \left[x - \frac{1}{2a} \sin 2ax \right]_{x_1}^{x_2}$$

$$P = \frac{A^2}{2} \left[(x_2 - x_1) - \frac{1}{2a} [\sin 2ax_2 - \sin 2ax_1] \right]$$

a. $x_1 = 25 - 5 = 20 \text{ pm}$, $x_2 = 25 + 5 = 30 \text{ pm}$

b. $x_1 = 50 - 5 = 45 \text{ pm}$, $x_2 = 50 + 5 = 55 \text{ pm}$

c. $x_1 = 90 - 5 = 85 \text{ pm}$, $x_2 = 90 + 5 = 95 \text{ pm}$

#16.

~~E₃~~ From Fig 39-4

$$E_3 = 280 \text{ eV}$$

$$U_0 = 450 \text{ eV}$$

$$\begin{aligned} KE &= U_0 - E_0 \\ &= 450 - 280 \\ &= 270 \text{ eV} \end{aligned}$$

$$\#20 \quad E_{n_x, n_y} = \frac{h^2}{8m} \left[\frac{n_x^2}{L_x^2} + \frac{n_y^2}{L_y^2} \right]$$

$$E_{n_x, n_y} = \frac{h^2}{8mL^2} \left[n_x^2 + \frac{n_y^2}{4} \right]$$

a. ground state $n_x = n_y = 1$

$$E_{1,1} = \frac{h^2}{8mL^2} \left[1 + \frac{1}{4} \right] = \frac{5h^2}{32mL^2}$$

b. 1st excited state is $n_x = 1, n_y = 2$

$$E_{1,2} = \frac{h^2}{4mL^2}$$

$$c. \quad E_{2,2} = \frac{h^2}{8mL^2} [4 + 1] = \frac{5h^2}{8mL^2}$$

2c - continue

d. Second Excited state

i) when,

$$n_x = 2, \quad n_y = 1$$

$$E_{21} = \frac{h^2}{8mL^2} \left[4 + \frac{1}{4} \right]$$

$$= \frac{17 h^2}{32 mL^2}$$

Third Excited-state

$$E_{22} = \frac{5 h^2}{8 mL^2}$$

$$\Delta E = E_{22} - E_{21}$$

$$= \frac{h^2}{8 mL^2} \left[5 - \frac{17}{4} \right]$$

$$= \frac{h^2}{8 mL^2} \left[\frac{20 - 17}{4} \right]$$

$$\Delta E = \frac{3 h^2}{32 mL^2}$$

#30.

$$P(r) = \frac{4}{a^3} r^2 e^{-2r/a}$$

a. $r=0$

$$P(0) = 0$$

b. $r=a$

$$P(a) = \frac{4}{a^3} a^2 e^{-2}$$

$$= \frac{4}{a} e^{-2}$$

c. $r=2a$

#35

~~$E = U + KE$~~ $KE = 13.6 \text{ eV}$

at $r=a$

$$U = \frac{e^2}{4\pi\epsilon_0 a} = \frac{(1.6 \times 10^{-19})^2}{4\pi(8.85 \times 10^{-12})(5.3 \times 10^{-11})}$$

$$= \frac{2.56 \times 10^{-38}}{5.89 \times 10^{-21}}$$

$$= 4.34 \times 10^{-18} \text{ J}$$

$$= -27.2 \text{ eV}$$

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$$\frac{1}{\lambda} = R \left[\frac{1}{n_{\text{low}}^2} - \frac{1}{n_{\text{high}}^2} \right]$$

$$\frac{1}{n_{\text{low}}^2} - \frac{1}{n_{\text{high}}^2} = \frac{1}{\lambda R}$$

$$= \frac{1}{102.6 \times 10^{-9} \times 1.097 \times 10^7}$$

$$\frac{1}{1} - \frac{1}{9} = 0.888$$

$$n_{\text{low}} = 1$$

$$n_{\text{high}} = 3$$

Lyman Series.