

This is not a replacement for your textbook. It is essential to study from your textbook!

Objectives

Ch 21: Coulomb's law

21-1 Coulomb's law

21-2 Charge is quantized

21-3 Charge is conserved

The symbol for **charge** is q

The SI unit for **charge** is the **coulomb** (C)

The smallest charge is that of the elementary charge e
 $e = 1.6 \times 10^{-19} \text{ C}$

All other charges are an **integer** multiple of e

$\Rightarrow$ charge is **quantized**.

$$q = 3e$$

possible

$$q = 3.5e$$

Not possible

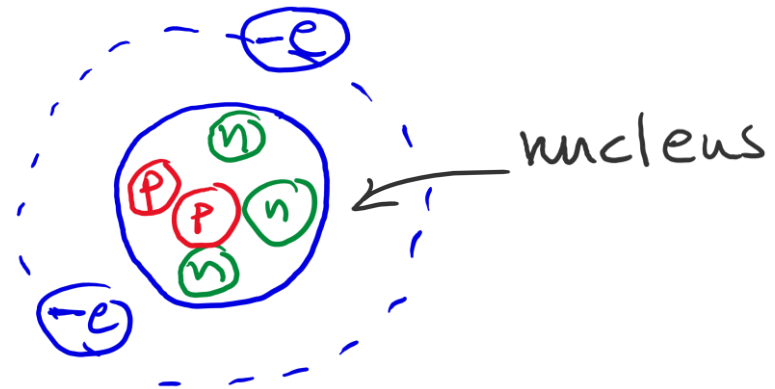
Charge can be **negative** or **positive**.

The total charge in an isolated system is **conserved**.

$$\text{positive charge} + \text{negative charge} = \text{constant}$$

	charge
electron	$-e = -1.6 \times 10^{-19} \text{ C}$
proton	$+e = +1.6 \times 10^{-19} \text{ C}$
neutron	0 has no charge

An atom is electrically neutral
 number of electrons = number of protons

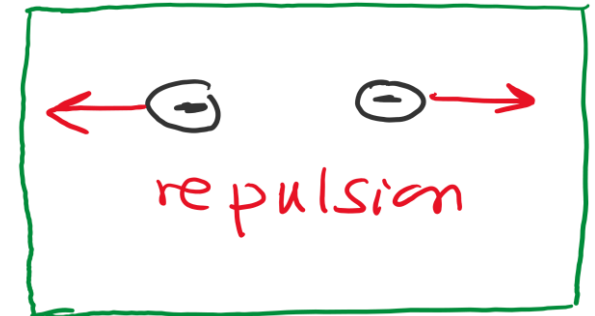
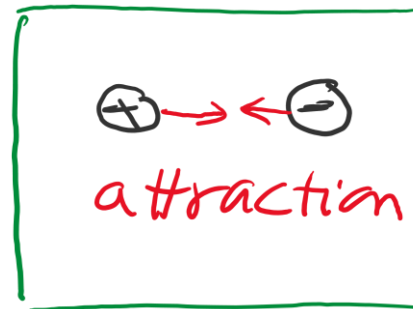
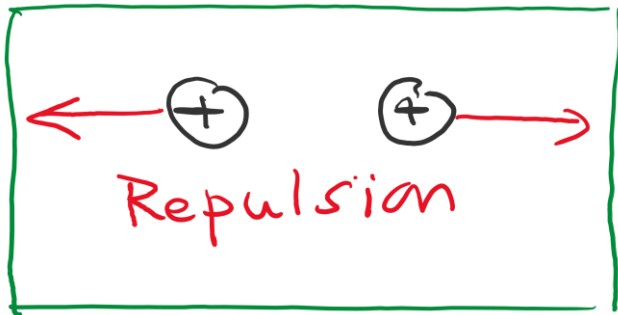
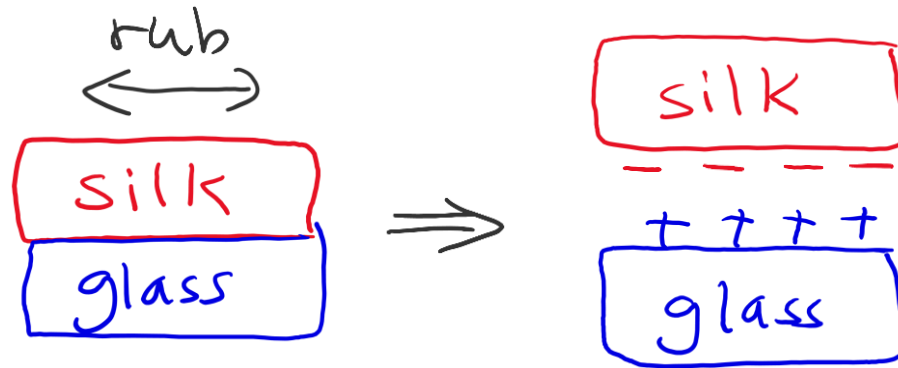


A positive ion = an atom that lost electrons

A negative ion = an atom that gained electrons

Electrostatic $\equiv$ charges are stationary
or moving very slowly.

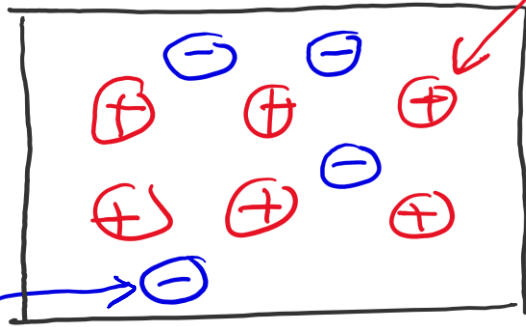
You can charge an object by rubbing



Conductors

Outermost electrons are loosely held by the nucleus

ions are fixed (solid)
move very slowly (liquid)



Outermost electrons can move freely

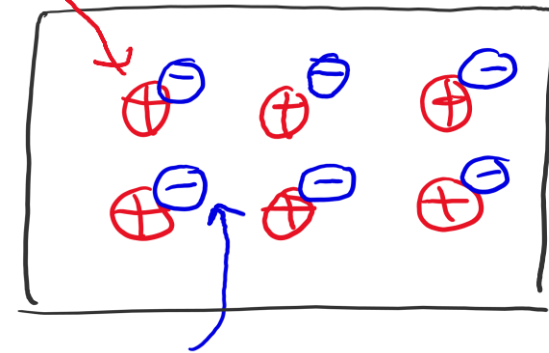
metal

tap water

human body

Insulators

Outermost electrons are tightly held by the nucleus



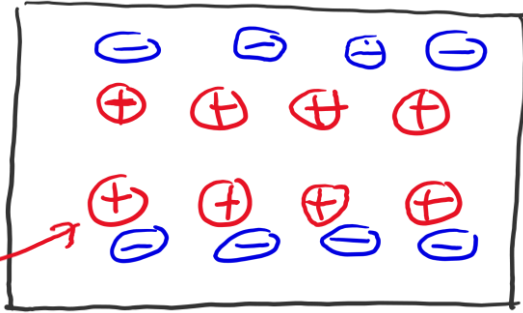
Outermost electrons cannot move

glass

pure water

plastic

neutral



ion are
fixed

Copper
(conductor)

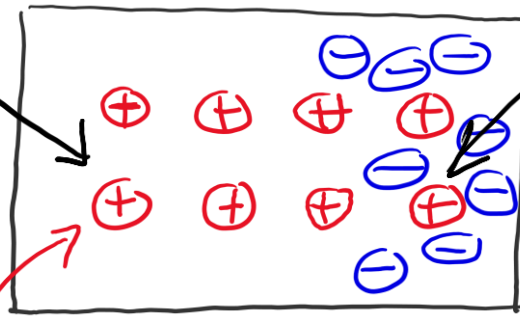
The conductor is neutral at all points.

positive charge density = negative charge density

lower electron density
≡ induced positive charge

higher electron density
≡ induced negative charge

neutral



ion are
fixed

Copper
(conductor)



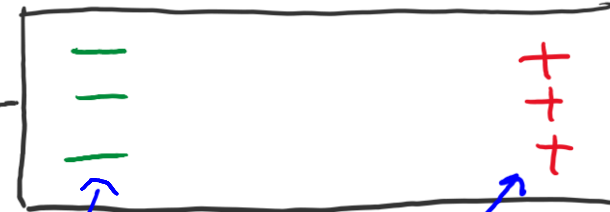
positively
charged
object

Charged
insulator



Attraction

Neutral conductor



induced
charge

Charged
insulator



Attraction

Neutral conductor



induced
charge

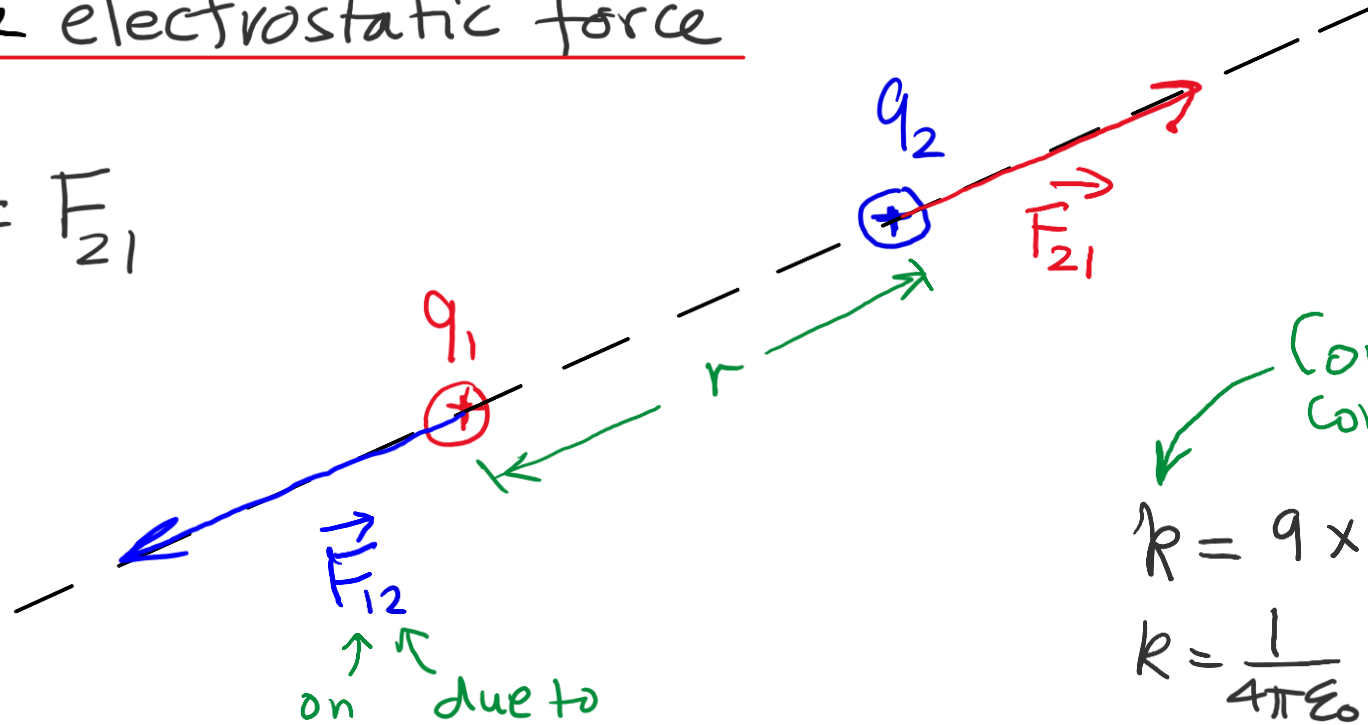
Coulomb's law

The magnitude of the electrostatic force

$$F_{12} = k \frac{|q_1||q_2|}{r^2} = F_{21}$$

on $\nearrow$ due to $\nwarrow$

$$\vec{F}_{12} = -\vec{F}_{21}$$



Coulomb constant

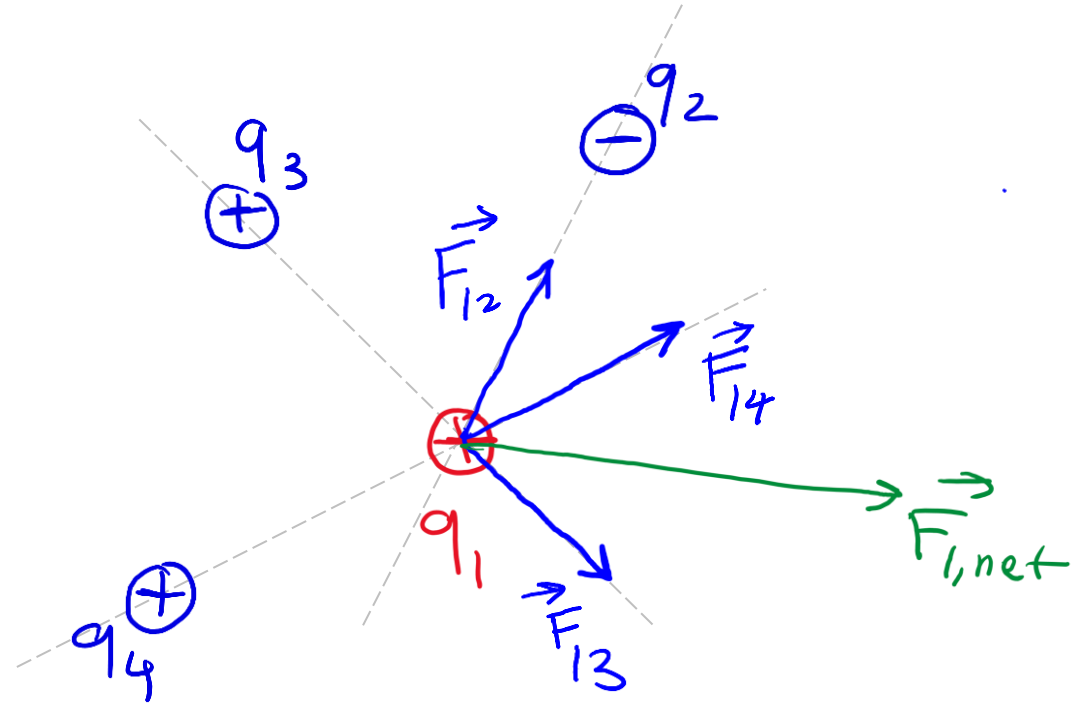
$$k = 9 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2}$$
$$k = \frac{1}{4\pi\epsilon_0}$$

permittivity constant

$$\epsilon_0 = 8.85 \times 10^{-12} \frac{\text{C}^2}{\text{Nm}^2}$$

The two forces form a third-law force pair.

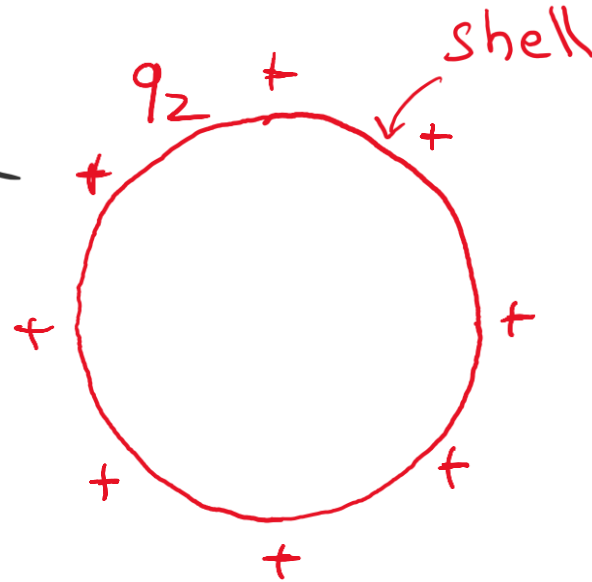
Principle of superposition



$$\vec{F}_{1,net} = \vec{F}_{12} + \vec{F}_{13} + \vec{F}_{14}$$

Shell theorem

charge q_2 is uniformly distributed over the surface



$\oplus q_1$

same as

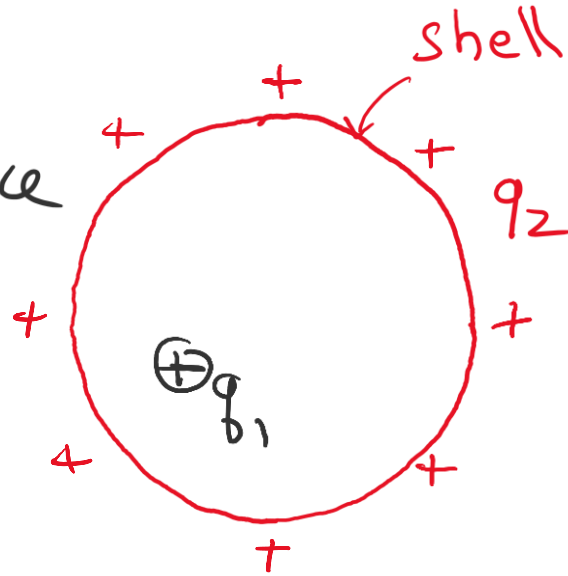
If q_1 is outside the shell,
you can replace the shell with a point
of charge q_2 at the center of the shell.

$\oplus q_2$

$\oplus q_1$

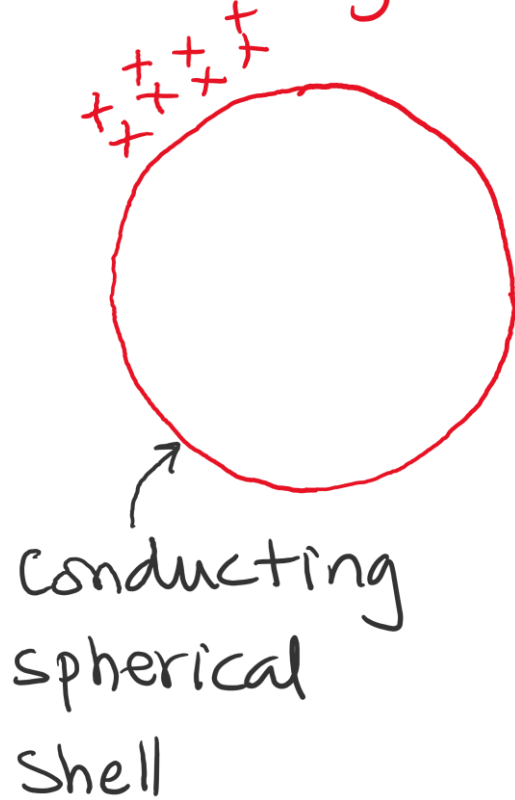
Shell theorem

charge q_2 is uniformly distributed over the surface

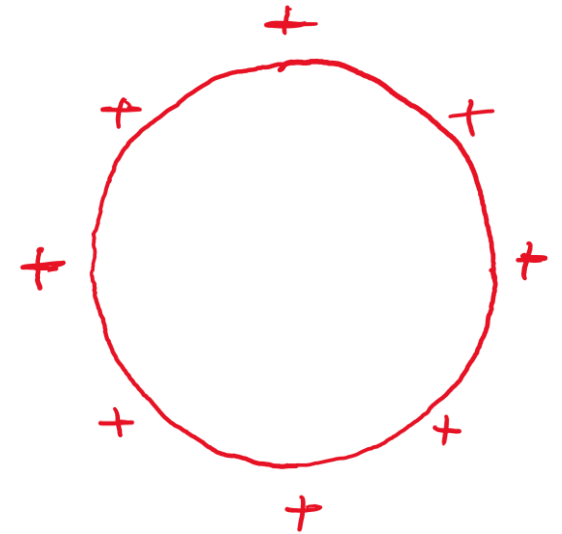


If q_1 is inside the shell,
the net force on q_1 due to the shell is zero.

excess charge



Free electrons in the conductor move to minimize the forces among charges



⇒ Excess charge on a spherical conductor spreads uniformly on the surface.

Example What is the electrostatic force on q_1 ?

$$\begin{aligned}q_1 &= 1 \mu\text{C} \\ q_2 &= 4 \mu\text{C} \\ q_3 &= -4 \mu\text{C}\end{aligned}$$

$$\vec{F}_{12} = k \frac{|q_1| |q_2|}{R^2} (-\hat{i}) = -9 \times 10^9 \frac{1 \times 10^{-6} \times 4 \times 10^{-6}}{(0.03)^2} \hat{i} = -(40 \text{ N}) \hat{i}$$

$$\vec{F}_{13} = k \frac{|q_1| |q_3|}{R^2} \{ \cos 60^\circ \hat{i} + \sin 60^\circ \hat{j} \} = (20 \text{ N}) \hat{i} + (35 \text{ N}) \hat{j}$$

$$\vec{F}_{1,\text{net}} = \vec{F}_{12} + \vec{F}_{13} = (-40 \text{ N} + 20 \text{ N}) \hat{i} + (35 \text{ N}) \hat{j} = -(20 \text{ N}) \hat{i} + (35 \text{ N}) \hat{j}$$

Magnitude of $\vec{F}_{1,\text{net}}$ $F_{1,\text{net}} = \sqrt{(-20)^2 + (35)^2}$

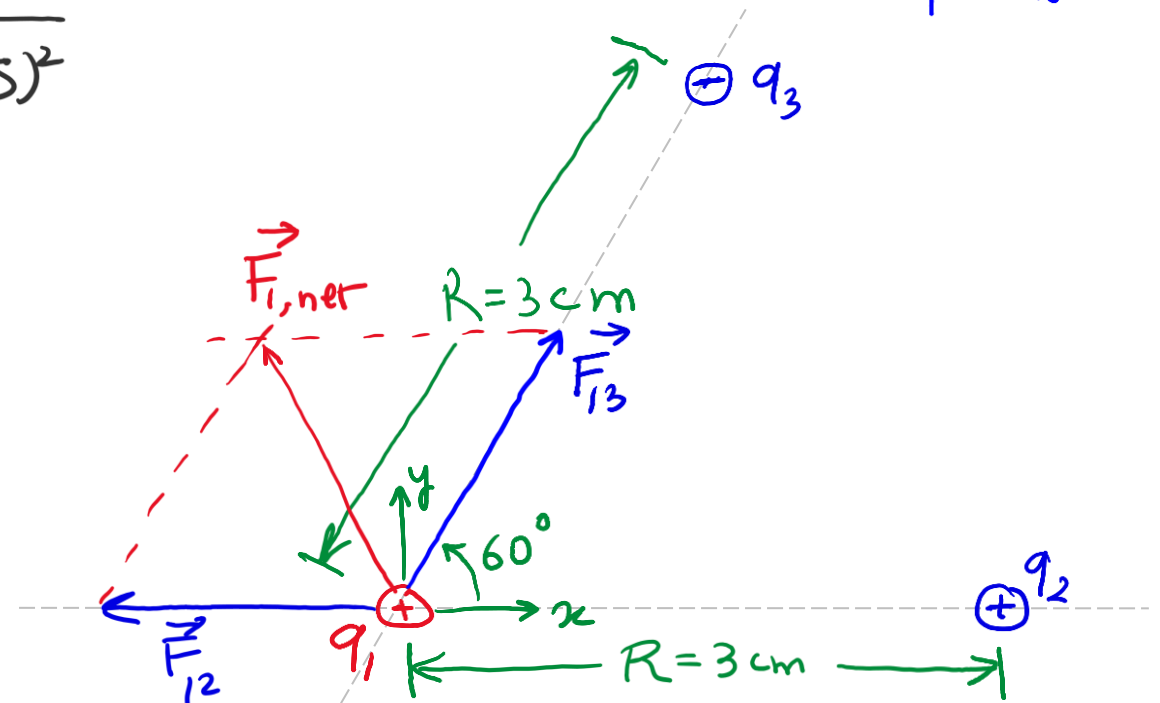
Angle of $\vec{F}_{1,\text{net}}$ from x-axis

$$\theta = \tan^{-1} \frac{35}{-20} = -60^\circ + 180^\circ = 120^\circ$$

When using $\tan^{-1}$, check the angle!

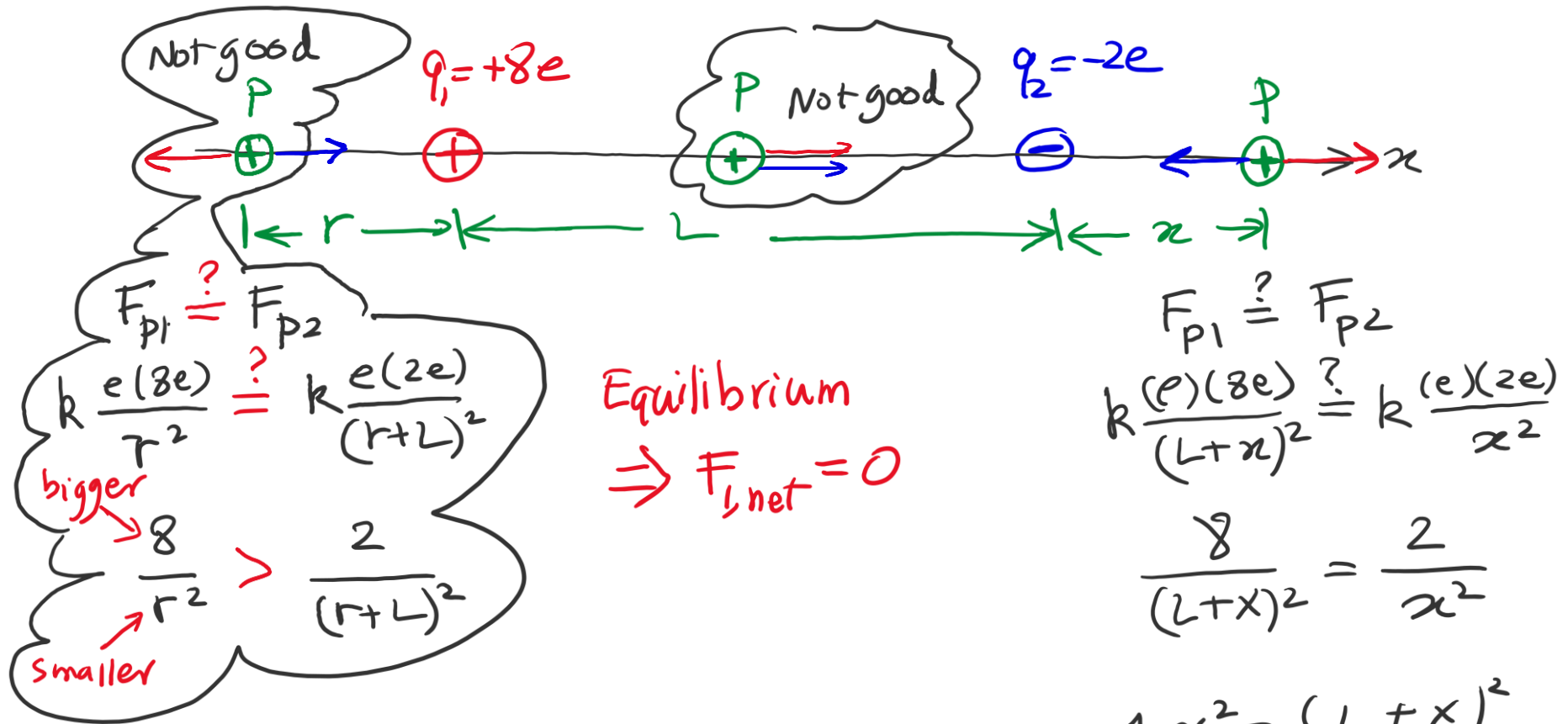
$\rightarrow \hat{i}$
unit vector

$\uparrow \hat{j}$
unit vector



Example

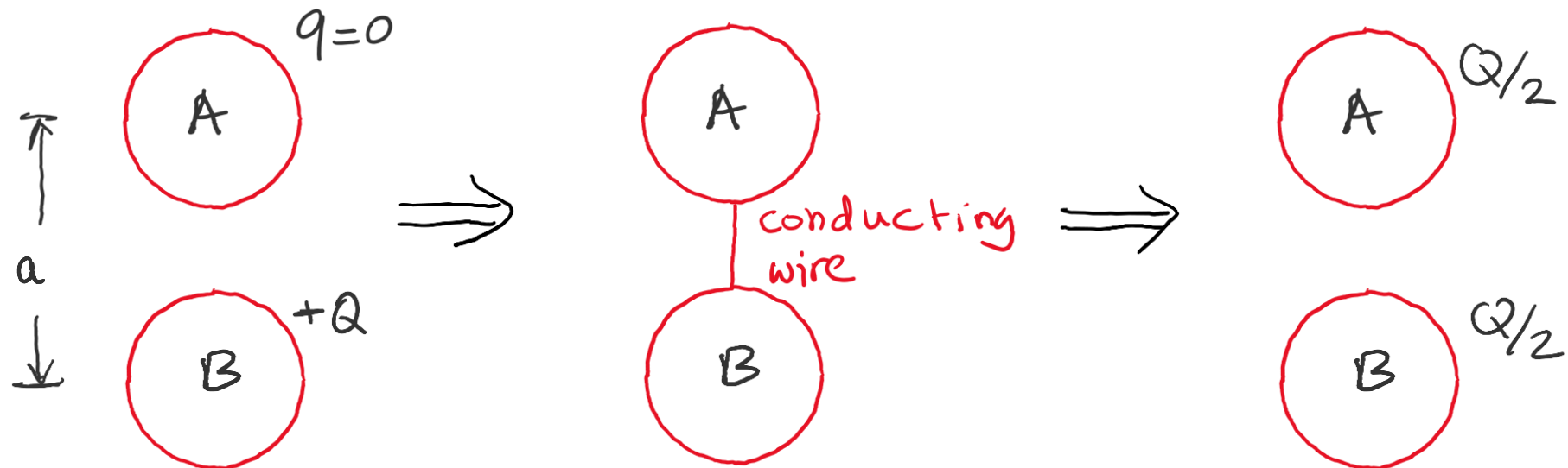
At what point on the x-axis a proton can be placed such that it is in equilibrium?



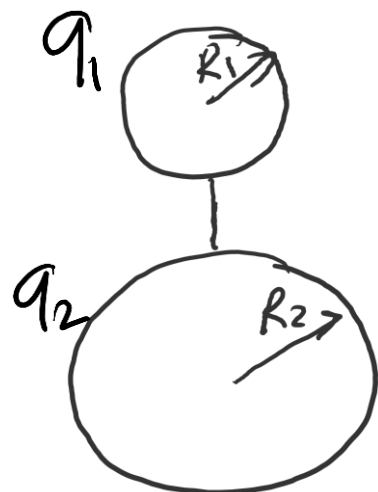
The equilibrium is not stable.

Example

Identical conducting spheres



$$F_{AB} = k \frac{|\frac{Q}{2}| |\frac{Q}{2}|}{a^2}$$



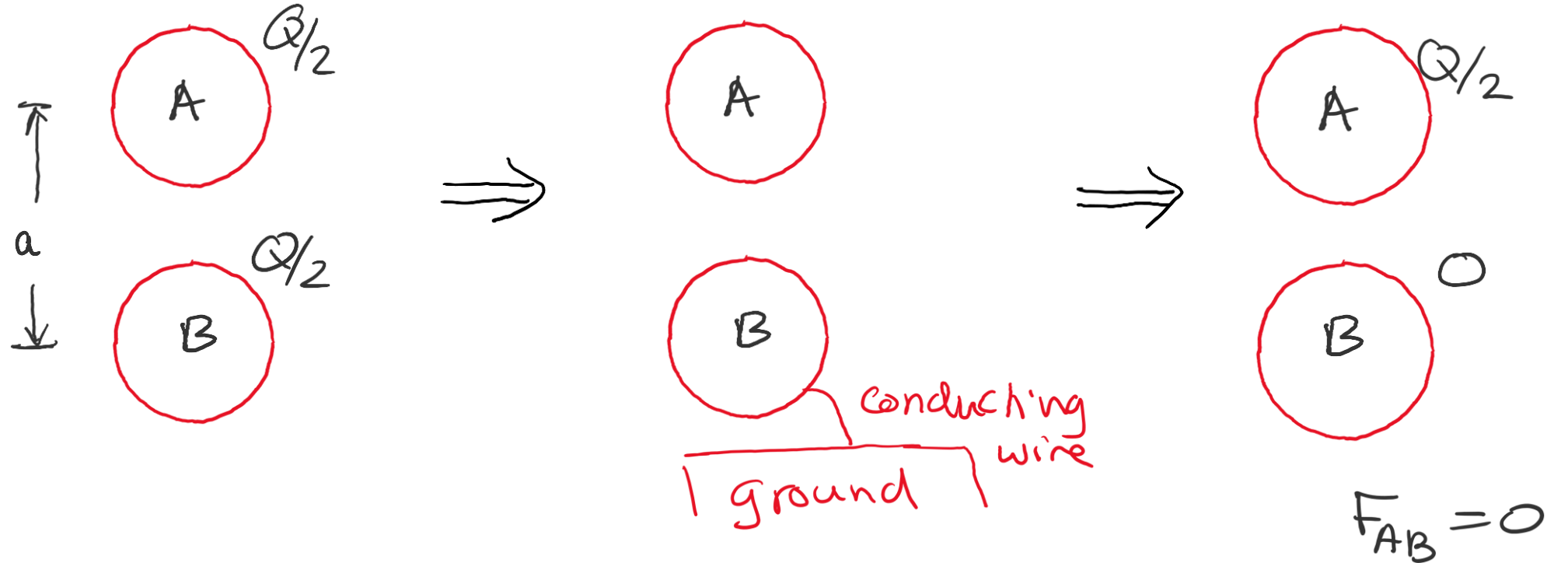
We will show later $\frac{q_1}{R_1} = \frac{q_2}{R_2} \xrightarrow{R_1=R_2} q_1 = q_2$

$$\frac{q_1}{R_1} = \frac{q_2}{R_2} \Rightarrow q_1 = \frac{R_1}{R_2} q_2$$

$$R_2 \gg R_1 \quad q_1 \approx 0$$

grounding

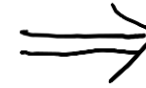
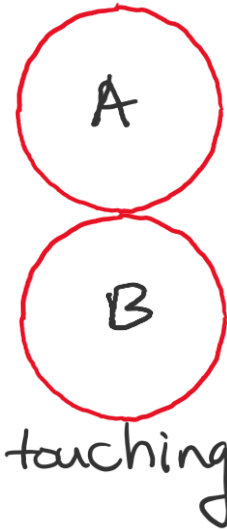
Example Identical conducting spheres



Question Identical conducting spheres

A $q_A = -50e$

B $q_B = +20e$



A $q_A = -15e$

B $q_B = -15e$