

Physics 102-Rec  
Quiz # 4  
Chapter 21

Date: 2 April 2002

Name: Key Id: \_\_\_\_\_ Sect: \_\_\_\_\_

1.

- (a) A Carnot refrigerator extract heat from the freezer maintained at  $-10^\circ\text{C}$  to the room at  $25^\circ\text{C}$ . What is the coefficient of performance of the refrigerator?

$$K = \frac{T_L}{T_H - T_L} = \frac{263}{35} = 7.5$$

$$\boxed{K = 7.5}$$

- (b) What is the amount of heat rejected to the room for 100 J of work is done on the refrigerator?

$$K = \frac{Q_L}{W} \Rightarrow Q_L = KW = 7.5 \times 100 = 750 \text{ J}$$

$$Q_H = Q_L + W = 750 + 100 = 850 \text{ J}$$

$$\boxed{Q_H = 850 \text{ J}}$$

2. A Carnot engine operates between two reservoirs maintained at  $300^\circ\text{C}$  and  $30^\circ\text{C}$ . If the heat gained the hot reservoir is 400 J for each cycle,

- (a) what is the heat ~~rejected~~ <sup>expelled</sup> by the engine?

$$\epsilon_c = 1 - \frac{T_L}{T_H} = 1 - \frac{303}{573} = 0.47$$

$$\frac{Q_L}{Q_H} = \frac{T_L}{T_H} \Rightarrow Q_L = Q_H \frac{T_L}{T_H} = 400 \frac{303}{573} = 212 \text{ J}$$

- (b) what is the work done by the engine?

$$\boxed{Q_L = 212 \text{ J}}$$

$$Q_H = Q_L + W \Rightarrow W = Q_H - Q_L = 400 - 212$$

$$\boxed{W = 188 \text{ J}}$$