Physics 102 Rec Quiz # 3 Chapter 19

Date: 19 March 2002

Name:	Key	Id:	Sect:

 A silver bar of length 30 cm and diameter 10 mm is used to transfer heat from a 100 °C reservoir to a 0 °C reservoir. How much heat is transferred per hour? (Thermal condictivity of silver = 427 W/m °C)

$$H = \frac{Q}{t} = \frac{X A \Delta T}{L}$$

$$\Rightarrow Q = H * t = \frac{X A \Delta T}{L} * t$$

$$= (427)(\pi r^{2})(100 - 0) \times 3600$$

$$= 427 \times \pi \times \alpha(5 \times 10^{3})^{2}(100) \times 3600$$

$$Q = 40244 J$$

2. Five moles of an ideal gas are compressed isothermally at 127 °C to half the initial volume. Find the work done by the gas during the process.

$$W = n RT ln(\frac{V_{f}}{V_{c}})$$

$$= 5 \times 8.31 \times 400 \times ln(\frac{1}{2}) = -11520 J$$

$$\overline{W = -11520 J}$$