Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 1

Q1.

A stretched string has a length of 2.00 m and a mass of 3.40 g. A transverse sinusoidal wave is travelling on this string, and is given by $y(x, t) = 0.030 \sin(0.75 x - 126 t)$, where x and y are in meters, and t is in seconds. What is the magnitude of the tension in this string?

A)	48	N
· •/		- '

- B) 60 N
- C) 17 N
- D) 29 N
- E) 35 N

Q2.

The average power of a sinusoidal wave on a stretched string is P. If an identical wave is sent simultaneously along the same string in the same direction but with a phase difference of 90° from the first wave, the new average power is

A) 2P

B) P

- C) 4P_
- D) $\sqrt{2}$ P
- E) P/2

Q3.

For a standing wave on a string fixed at both ends

- A) the midpoint is an antinode for odd harmonics.
- B) the midpoint is an antinode for even harmonics.
- C) the midpoint is a node for odd harmonics.
- D) the shortest wavelength corresponds to the fundamental mode.
- E) the amplitude of all points on the string is the same.

Q4.

A string that is stretched between fixed supports oscillates in a third-harmonic standing wave pattern. The displacement of the wave is given by $y(x,t) = (0.10) \sin (\pi x/5) \cos (12\pi t)$,

where *x* and *y* are in meters, and *t* is in seconds. What is the length of the string?

- A) 15 m
- B) 6.7 m
- C) 10 m
- D) 60 m

E) 25 m

Q5.

A string that is stretched between fixed supports has resonant frequencies of 385 and 430 Hz, with no intermediate resonant frequencies. What is the frequency of the seventh harmonic?

- A) 315 Hz
- B) 45 Hz
- C) 2700 Hz
- D) 655 Hz
- E) 3010 Hz

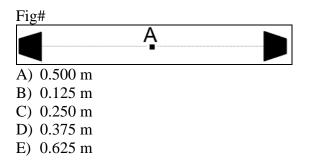
Q6.

If the intensity of a sound wave traveling in air with constant frequency is doubled, then

- A) the wave speed remains the same.
- B) the displacement amplitude remains the same.
- C) the displacement amplitude is doubled.
- D) the sound level is doubled.
- E) the displacement amplitude is halved.

Q7.

Two speakers, separated by 2.00 m, face each other as shown in **Figure 1**. They are driven by the same generator, and emit sound waves with a frequency of 170 Hz, that are initially in phase. A listener is initially at point **A**, which is at the midpoint between the two speakers. What is the shortest distance he should move to find a point of destructive interference? [Take the speed of sound to be 340 m/s]



Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 3

Q8.

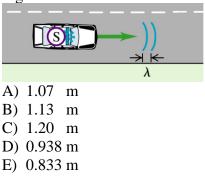
A tube open at both ends has length L_A . A tube open only at one end has length L_B . If the two tubes have the same fundamental frequency, then

A) $L_A = 2L_B$ B) $L_A = L_B/2$ C) $L_A = L_B/4$ D) $L_A = L_B$ E) $L_A = 4L_B$

Q9.

A police car, moving at 20.0 m/s, emits a sound wave with a frequency of 300 Hz. Find the wavelength of the sound wave in front of the car, as shown in **Figure 2**. [Take the speed of sound in air to be 340 m/s]

Fig#



Q10.

The melting point of sulfur is 444.6 °C and is 586.1 F° below its boiling point. Determine the boiling point of sulfur in degrees Celsius.

A) 770.2 °C B) 118.0 °C C) 1031 °C D) 1500 °C E) 214.2 °C

Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 4

Q11.

An iron tank is completely filled with 2.80 m³ of water when both the tank and the water are at a temperature of 32.0 °C. When the tank and the water have cooled to 18.0 °C, what additional volume of water can be put into the tank? [$\alpha_{iron} = 12.0 \times 10^{-6}$ / C°, $\beta_{water} = 4.79 \times 10^{-4}$ / C°]

A) $17.4 \times 10^{-3} \text{ m}^3$ B) $1.41 \times 10^{-3} \text{ m}^3$ C) $18.8 \times 10^{-3} \text{ m}^3$ D) $0.470 \times 10^{-3} \text{ m}^3$ E) $18.3 \times 10^{-3} \text{ m}^3$

Q12.

A 100-g ice cube at 0.0 °C is placed in 650 g of water at 18 °C. If the system is isolated, what is the final temperature?

A) 5.0 °C
B) 0.0 °C
C) 22 °C
D) 28 °C
E) 12 °C

Q13.

A copper rod has a length of 60 cm. One end is maintained at 80 $^{\circ}$ C and the other end is at 20 $^{\circ}$ C. In steady state, what is the temperature of the rod at a point which is 20 cm from the hot end?

 $[k_{copper} = 401 \text{ W/m.K}]$

A) 60 °C

- B) $40 \degree C$
- C) 35 °C
- D) 25 °C
- E) 50 °C

Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 5

Q14.

A 5 moles of an ideal gas expand isobarically from $T_i = 25$ °C to $T_f = 75$ °C. Calculate the work done by the gas during this process.

A) $2.1 \times 10^3 \text{ J}$ B) $4.5 \times 10^3 \text{ J}$ C) $1.2 \times 10^3 \text{ J}$ D) $5.4 \times 10^3 \text{ J}$ E) zero

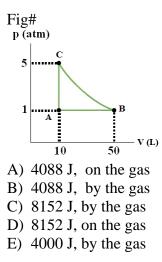
Q15.

An ideal gas has a density of 3.75 kg/m^3 and is at a pressure of 1.00 atm. Determine the rms speed of the molecules of this gas.

A) 284 m/s
B) 94.9 m/s
C) 164 m/s
D) 1070 m/s
E) 616 m/s

Q16.

An ideal monatomic gas is taken through cycle $A \rightarrow B \rightarrow C \rightarrow A$, shown in the *p*-*V* diagram of **Figure 3**, where process $B \rightarrow C$ is isothermal. Calculate the net work done in one cycle.



Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 6

Q17.

One mole of an ideal monatomic gas is initially at a pressure of 1.01×10^5 Pa, a temperature of 300 K, and has a volume of 1.00 L. It is compressed adiabatically to a volume of 0.0667 L. Calculate the magnitude of the work done during this process.

A) 19.0 kJ

- B) 6.75 kJ
- C) 94.3 kJ
- D) 2.49 kJ
- E) Zero

Q18.

A system consists of two large thermal reservoirs in contact with each other, one at a temperature of 300 °C and the other at a temperature 200 °C. If 600 J of heat is transferred from the 300 °C reservoir to the 200 °C reservoir, what is the change in entropy of this system?

A) 0.221 J/K
B) 1.00 J/K
C) 5.00 J/K
D) -1.00 J/K
E) 2.31 J/K

Q19.

A Carnot refrigerator is operated between two heat reservoirs at temperatures of 320 K and 270 K. In each cycle, the refrigerator extracts 415 J of heat from the cold reservoir. If the refrigerator completes 165 cycles each minute, what is the power input required to operate it?

A) 211 W
B) 178 W
C) 815 W
D) 224 W
E) 317 W

Phys102	First Major-103	Zero Version
	Saturday, July 16, 2011	Page: 7

Q20.

Which of the processes on an ideal gas shown in **Figure 4** results in the minimum change in entropy of the gas in changing the gas from state S to State F?



