

Questions  
Chapter 1  
Measurement

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# 1-3 Changing Units

## M1-042

Express speed of sound, 330 m/s in miles/h  
(1 mile = 1609 m )

- A) 147 miles/h
- B) 330 miles/h
- C) 738 miles/h
- D) 0.205 miles/h
- E) 980 miles/h

Answer C

# 1-3 Changing Units

M1-041

1 shake =  $10^{-8}$  seconds. Find out how many nano seconds (ns) are there in 1 shake.

(1 nano =  $10^{-9}$ )

- A) 0.1 ns
- B) 0.01 ns
- C) 100 ns
- D) 0.001 ns
- E) 10 ns

Answer E

# 1-3 Changing Units

## M1-991

How many molecules of water are there in a cup containing 250 cm<sup>3</sup> of water?

Molecular mass of H<sub>2</sub>O = 18 g/mole

Density of water = 1.0 g/cm<sup>3</sup>

Avogadro s number = 6.02 x 10<sup>23</sup> molecules/mole

- A) 6.0 x 10<sup>23</sup>
- B) 8.4 x 10<sup>24</sup>
- C) 1.9 x 10<sup>26</sup>
- D) 3.7 x 10<sup>28</sup>
- E) 2.5 x 10<sup>3</sup>

Answer B

## 1-3 Changing Units

M1-991

Using the fact that the speed of light in space is about  $3.00 \times 10^8$  m/s, determine how many miles light will travel in one hour.

(1 mile = 1.61 km)

- A)  $6.71 \times 10^8$  miles
- B)  $2.50 \times 10^6$  miles
- C)  $5.40 \times 10^9$  miles
- D)  $8.32 \times 10^3$  miles
- E)  $4.83 \times 10^2$  miles

Answer A

# 1-4 Dimensional Analysis

M1-062

Suppose

$$A = \frac{B^n}{C^m}$$

where  $A$  has dimensions  $LT$ ,  $B$  has dimensions  $L^2T^{-1}$ , and  $C$  has dimensions  $LT^2$ . Then the exponents  $n$  and  $m$  have the values:

- A)  $n = 1/5$  ;  $m = 3/5$
- B)  $n = 2$  ;  $m = 3$
- C)  $n = 4/5$  ;  $m = -1/5$
- D)  $n = 1/5$  ;  $m = -3/5$
- E)  $n = 1/2$  ;  $m = 1/2$

Answer D

## 1-4 Dimensional Analysis

M1-071

The position  $x$  of a particle is given by:  $x = B t^2 + \frac{C}{B} t$

where  $x$  is in meters and  $t$  is in seconds. The dimension of  $C$  is:

A)  $\frac{L^2}{T^3}$

B)  $\frac{L}{T}$

C)  $L$

D)  $T$

E)  $\frac{L}{T^2}$

Answer A

## 1-7 Density

### M1-062

From the fact that the average density of the Earth is  $5.50 \text{ g/cm}^3$  and its mean radius is  $6.37 \times 10^6 \text{ m}$ , the mass of the Earth is:

- A)  $7.01 \times 10^{17} \text{ kg}$
- B)  $3.98 \times 10^{21} \text{ kg}$
- C)  $5.95 \times 10^{24} \text{ kg}$
- D)  $2.80 \times 10^{18} \text{ kg}$
- E)  $5.50 \times 10^{23} \text{ kg}$

Answer C



## 1-7 Density

### M1-061

An aluminum cylinder of density  $2.70 \text{ g/cm}^3$ , a radius of  $2.30 \text{ cm}$ , and a height of  $1.40 \text{ m}$  has the mass of:

- A)  $25.0 \text{ kg}$
- B)  $45.1 \text{ kg}$
- C)  $13.8 \text{ kg}$
- D)  $8.50 \text{ kg}$
- E)  $6.28 \text{ kg}$

Answer E

## 1-7 Density

### M1-052

A nucleus of volume  $3.4 \times 10^3 \text{ fm}^3$  and mass of  $1.0 \times 10^2 \text{ u}$  has a density of:

( $1 \text{ fm} = 10^{-15} \text{ m}$ ,  $1 \text{ u} = 1.7 \times 10^{-27} \text{ kg}$ )

- A)  $5.0 \times 10^{16} \text{ kg/m}^3$
- B)  $1.0 \times 10^3 \text{ kg/m}^3$
- C)  $3.4 \times 10^{14} \text{ kg/m}^3$
- D)  $12 \times 10^3 \text{ kg/m}^3$
- E)  $3.6 \times 10^{13} \text{ kg/m}^3$

Answer A

## 1-7 Density

M1-042

A cylindrical can, 6.00 inches high and 3.00 inches in diameter is filled with water. Density of water is  $1.00 \text{ g/cm}^3$ . What is the mass of water in the can in gram ?  
(1 inch = 2.54 cm) .

- A) 277 g
- B) 695 g
- C) 182 g
- D) 107 g
- E) 2780 g

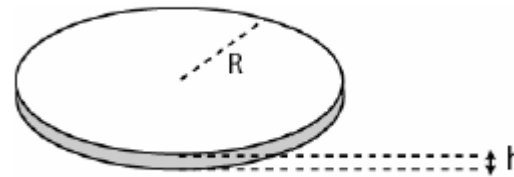
Answer B

# 1-7 Density

## M1-041

A drop of oil (mass = 0.90 milligram and density =  $918 \text{ kg/m}^3$ ) spreads out on a surface and forms a circular thin film of radius = 41.8 cm and thickness  $h$ . Find  $h$  in nano meter (nm). (1 nano =  $10^{-9}$ )

- A) 0.60 nm
- B) 0.00060 nm
- C) 0.15 nm
- D) 1.8 nm
- E) 0.030 nm



Answer D