

## MAJOR 2 (002)

- Q0  
(1)Q0 What is the change in entropy of 200-g of water as its  
002Q0 temperature increases from 0 degrees Celsius to  
21Q0 50 degrees Celsius. [For water: the specific heat =  
Q0 4.19 kJ/(kg.K) and the latent heat of fusion = 333 kJ/kg.]  
Q0  
A1  $1.41 \cdot 10^{**2}$  J/K.  
A2  $0.35 \cdot 10^{**3}$  J/K.  
A3  $4.19 \cdot 10^{**3}$  J/K.  
A4  $3.35 \cdot 10^{**3}$  J/K.  
A5  $2.55 \cdot 10^{**3}$  J/K.  
Q0  
(2)Q0 An ideal engine absorbs heat at 527 degrees Celsius  
002Q0 and rejects heat at 127 degrees Celsius. If it has to  
21Q0 produce useful mechanical work at the rate of 750 Watts,  
Q0 it must absorb heat at the rate of:  
Q0  
A1 1500 Watts.  
A2 750 Watts.  
A3 2250 Watts.  
A4 527 Watts.  
A5 375 Watts.  
Q0  
(3)Q0 A heat engine has a monatomic gas as the working substance and  
21 Q0 its operating cycle is shown by the P-V diagram in Figure 1.  
992Q0 In one cycle, 18.2 kJ of heat energy is absorbed by the  
002Q0 engine. Find the efficiency of the heat engine.  
Q0  
A1 0.44  
A2 0.55  
A3 0.31  
A4 0.25  
A5 0.22  
Q0  
(4)Q0 A negative charge is placed at the center of a square. Each  
002Q0 corner of the square has a fixed charge of  $1.00 \cdot 10^{**(-6)}$  C.  
22 Q0 If the resulting force acting on each charge is zero,  
Q0 the magnitude of the negative charge is:  
Q0  
A1  $0.96 \cdot 10^{**(-6)}$  C.  
A2  $9.60 \cdot 10^{**(-6)}$  C.  
A3  $6.92 \cdot 10^{**(-6)}$  C.  
A4  $0.69 \cdot 10^{**(-6)}$  C.  
A5  $0.77 \cdot 10^{**(-6)}$  C.  
Q0  
(5)Q0 Two neutral metal sphere are separated by 0.3 km. How much  
002Q0 electric charge must be transferred from one sphere to the  
22 Q0 other so that their electrical attraction is  $10^{**3}$  N?  
Q0  
A1 0.1 C.  
A2 0.2 C.  
A3 0.4 C.  
A4 0.6 C.  
A5 0.9 C.  
Q0  
(6)Q0 A point charge of 4.0 nano-C is located at a point having  
23 Q0 coordinates (30.0 cm, 40.0 cm). At what point will the  
992Q0 electric field be 72 N/C and pointing in the negative  
002Q0 y-direction?  
Q0  
A1 (30.0, -30.7) cm

A2 (30.0, 49.9) cm  
 A3 (30.0, 70.7) cm  
 A4 (30.0, -49.9) cm  
 A5 (10.0, -89.9) cm  
 Q0  
 (7)Q0 An electric dipole consists of a positive charge of magnitude  
 002Q0  $6.0 \times 10^{(-6)}$  C at the origin and a negative charge of magnitude  
 002Q0  $6.0 \times 10^{(-6)}$  C on the x-axis at  $x = 3.0 \times 10^{(-3)}$  m.  
 23 Q0 Its dipole moment is:  
 Q0  
 A1  $1.8 \times 10^{(-8)}$  C.m, in the negative x direction.  
 A2  $1.8 \times 10^{(-8)}$  C.m, in the positive x direction.  
 A3 Zero because the net charge is Zero.  
 A4  $1.8 \times 10^{(-8)}$  C.m, perpendicular to the x-axis.  
 A5  $3.6 \times 10^{(-8)}$  C.m, in the negative x direction.  
 Q0  
 (8)Q0 A charged particle has a mass of  $2.0 \times 10^{(-4)}$  kg. If it is  
 002Q0 held stationary by a downward 300 N/C electric field, the  
 23 Q0 charge of the particle is:  
 Q0  
 A1  $-6.5 \times 10^{(-6)}$  C.  
 A2  $6.5 \times 10^{(-6)}$  C.  
 A3  $-1.5 \times 10^{(-6)}$  C.  
 A4  $1.5 \times 10^{(-6)}$  C.  
 A5  $-3.0 \times 10^{(-6)}$  C.  
 Q0  
 09 Q0 Two uniformly charged, concentric and hollow, spheres have  
 24 Q0 radii  $r$  and  $1.5r$ . The charge of the inner sphere is  $q/2$  and  
 002Q0 that on the outer sphere is  $3q/2$ . Find the electric field at  
 Q0 a distance  $2.0r$  from the center of the spheres.  
 Q0  
 A1  $0.5kq/(r^2)$ .  
 A2  $0.13kq/(r^2)$ .  
 A3  $0.25kq/(r^2)$ .  
 A4  $0.35kq/(r^2)$ .  
 A5 Zero.  
 Q0  
 10 Q0 An infinitely long line has a charge density of 7.6 nano-C/m.  
 24 Q0 Calculate the electric flux through a spherical surface of  
 992Q0 radius  $R = 7.7$  cm whose center, C, lies on the line charge as  
 002Q0 shown in Figure 3.  
 Q0  
 A1 132  $(N \cdot m^2)/C$ .  
 A2 415  $(N \cdot m^2)/C$ .  
 A3 610  $(N \cdot m^2)/C$ .  
 A4 92.0  $(N \cdot m^2)/C$ .  
 A5 Zero.  
 Q0  
 11 Q0 Fig. 7 shows two parallel plates, infinite and non-conducting,  
 24 Q0 with surface charge densities of  $8.9 \times 10^{(-4)}$  C/m<sup>2</sup> and  
 002Q0  $-8.9 \times 10^{(-4)}$  C/m<sup>2</sup>. B, a ball with negligible mass, carries  
 Q0 a positive charge of  $6.0 \times 10^{(-8)}$  C and is attached to point A  
 Q0 with a non-conducting string of length 10 cm. At equilibrium,  
 Q0 the tension in the string is:  
 Q0  
 A1 6.0 N.  
 A2 1.5 N.  
 A3 3.0 N.  
 A4 0.3 N.  
 A5 Zero.  
 Q0

002Q0

- A1 257 degrees Celsius.
- A2 531 degrees Celsius.
- A3 340 degrees Celsius.
- A4 321 degrees Celsius.
- A5 132 degrees Celsius.

Q0

12Q0 A 100 g of water at 100 degrees Celsius is added to a 20-g aluminum cup containing 50 g of water at 20 degrees Celsius.

19 Q0 What is the equilibrium temperature of the system?

002Q0 The specific heat of aluminum is  $900 \text{ J/(kg}\cdot\text{K)}$  and the specific heat of water is  $4186 \text{ J/(kg}\cdot\text{K)}$ .

Q0

- A1 72 degrees Celsius.
- A2 63 degrees Celsius.
- A3 14 degrees Celsius.
- A4 55 degrees Celsius.
- A5 95 degrees Celsius.

Q0

13 Q0 A solid aluminum rod, of length 1.60 m and cross-sectional area

19 Q0 of  $3.14 \cdot 10^{-4} \text{ m}^2$ , has one end in boiling water and the

002Q0 other end in ice. How much ice melts in one minute?

Q0 [The thermal conductivity of aluminum is  $205 \text{ Watts/(m}\cdot\text{K)}$

Q0 and the heat of fusion of water is  $3.35 \cdot 10^5 \text{ J/kg}$ .]

Q0 (neglect any heat loss, by the system, to the surrounding)

Q0

- A1  $7.2 \cdot 10^{-4} \text{ kg}$ .
- A2  $7.9 \cdot 10^{-2} \text{ kg}$ .
- A3  $6.3 \cdot 10^{-4} \text{ kg}$ .
- A4  $5.8 \cdot 10^{-4} \text{ kg}$ .
- A5  $3.2 \cdot 10^{-3} \text{ kg}$ .

Q0

14Q0 An iron ball has a diameter of 6.0 cm and is 0.01 mm too large

19 Q0 to pass through a hole in a brass ring when both are at a

002Q0 temperature of 30 degrees Celsius. To what temperature should

Q0 the brass ring be heated so that the ball just passes through

Q0 the hole? [The coefficient of volume expansion of

Q0 iron =  $3.6 \cdot 10^{-5} \text{ K}^{-1}$  and of brass =  $5.7 \cdot 10^{-5} \text{ K}^{-1}$ ]

Q0

- A1 39 degrees Celsius.
- A2 59 degrees Celsius.
- A3 47 degrees Celsius.
- A4 52 degrees Celsius.
- A5 32 degrees Celsius.

Q0

15Q0 5 moles of hydrogen gas occupy a balloon that is inflated to a

20 Q0 volume of  $0.3 \text{ m}^3$  and at 1.0 atmospheric pressure. What is the

002Q0 root-mean square velocity of the molecules inside the balloon?

Q0 [The mass of hydrogen atom is  $1.66 \cdot 10^{-27} \text{ kg}$ .]

Q0

- A1  $4.3 \cdot 10^3 \text{ m/s}$ .
- A2  $3.4 \cdot 10^2 \text{ m/s}$ .
- A3  $3.0 \cdot 10^9 \text{ m/s}$ .
- A4  $2.2 \cdot 10^3 \text{ m/s}$ .
- A5  $1.3 \cdot 10^3 \text{ m/s}$ .

Q0

16 Q0 For an ideal gas, which of the following statements is FALSE:

002Q0 is charged by a battery to a potential difference of 12.0 volts.  
26 Q0 The charging battery is then disconnected and oil with  
Q0 dielectric constant = 4.0 fills the inside space between the  
Q0 plates. The resulting potential difference, in volts, between  
Q0 the plates is:  
Q0  
A1 3.  
A2 12.  
A3 48.  
A4  $1.0 \cdot 10^{(-9)}$ .  
A5  $3.0 \cdot 10^{(-9)}$ .  
Q0  
18 Q0 If  $V_{ab}$  is equal to 50 V, find the charge stored and the  
26 Q0 potential difference across the 25 micro-F capacitor shown  
991Q0 in Figure 5.  
002Q0  
A1 250 micro-C and 10 V.  
A2 300 micro-C and 20 V.  
A3 600 micro-C and 10 V.  
A4 600 micro-C and 20 V.  
A5 250 micro-C and 40 V.  
Q0  
19 Q0 If 110 Volts is applied to a wire, the current density is  
27 Q0  $1.5 \cdot 10^{*6}$  A/m\*\*2. If the resistivity of the wire is  
002Q0  $48.2 \cdot 10^{*(-8)}$  Ohm.m, the length of the wire is:  
Q0  
A1 152 m.  
A2 76 m.  
A3 254 m.  
A4 38 m.  
A5 19 m.  
Q0  
20 Q0 At what temperature would the resistance of a conductor be  
27 Q0 double its resistance at 30 degrees Celsius?  
002Q0 [The temperature coefficient of resistivity of the conductor  
Q0 is  $2.0 \cdot 10^{*(-2)}$  K\*\*(-1)]  
Q0  
A1 80 degrees Celsius.  
A2 -20 degrees Celsius.  
A3 20 degrees Celsius.  
A4 50 degrees Celsius.  
A5 60 degrees Celsius.

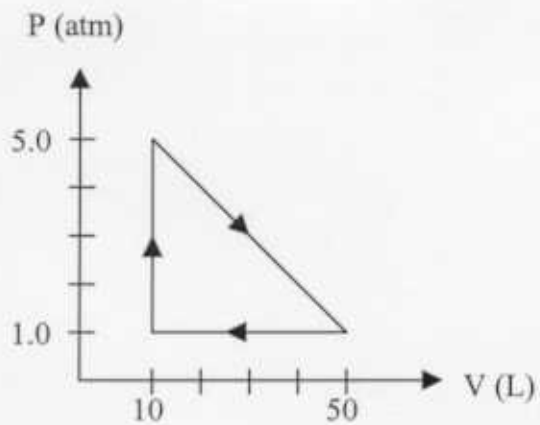


FIGURE 1

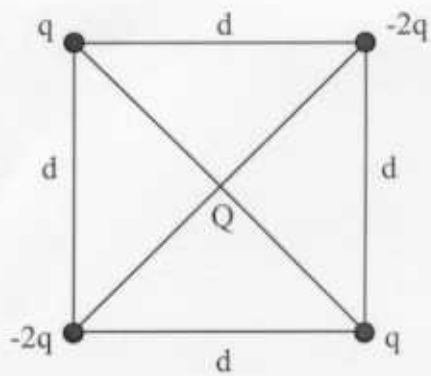


FIGURE 2

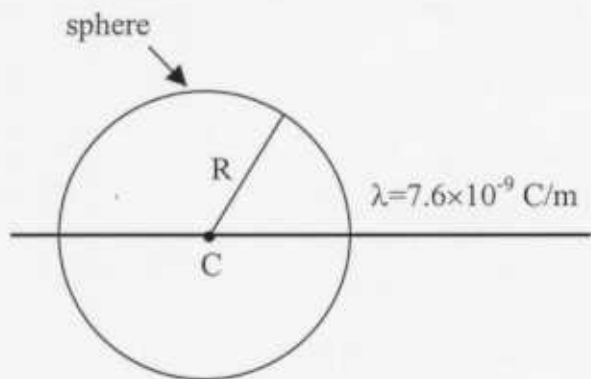


FIGURE 3

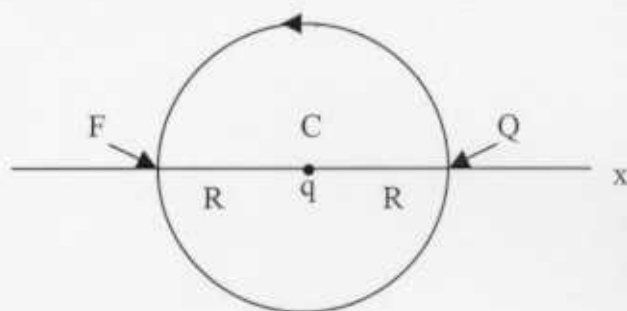


FIGURE 4

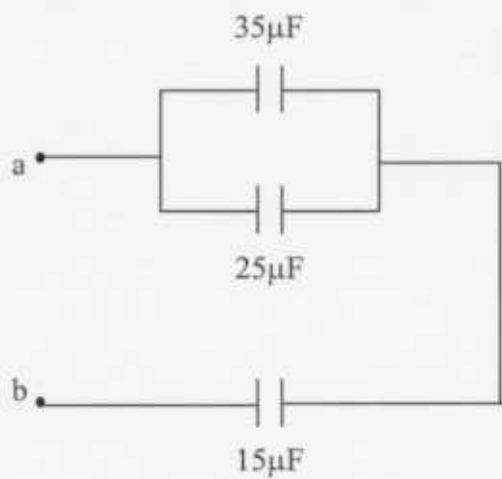


FIGURE 5

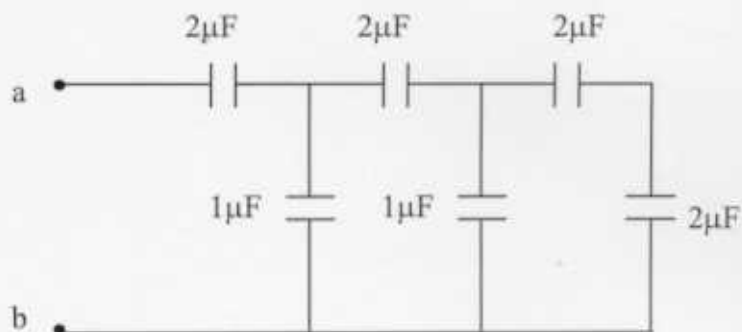


FIGURE 6

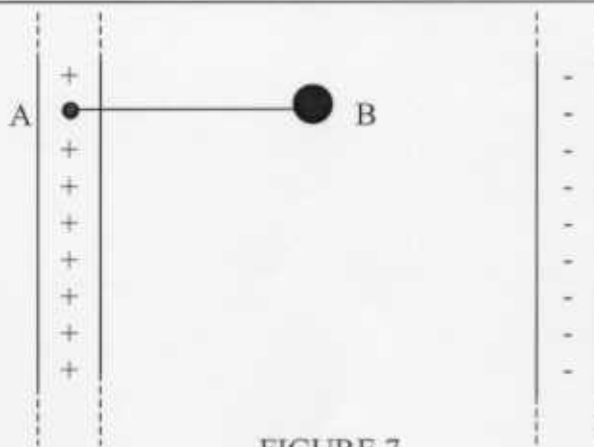


FIGURE 7