

Solutions: H.W. CH. # 1; Fall Semester 2001 (011)

5. Given:  $R = 6.37 \times 10^6 \text{ m}$ ; Assume Earth is sphere

Its circumference in Kilometers

$$\text{Circumference} = 2\pi R = 2\pi (6.37 \times 10^6 \text{ m}) \left( \frac{1 \text{ km}}{10^3 \text{ m}} \right) = 40.04 \times 10^3 \text{ km}$$

We have to get the answer in 3 significant figures:

$$\underline{\underline{4.00 \times 10^4 \text{ km}}}$$

b) Surface Area =  $4\pi R^2 = 4\pi (6.37 \times 10^6 \text{ m})^2 \left( \frac{1 \text{ km}}{10^3 \text{ m}} \right)^2$   
 $4\pi (6.37)^2 \times 10^6 \text{ km}^2 = \underline{\underline{5.10 \times 10^8 \text{ km}^2}}$

c) Volume =  $\frac{4}{3}\pi R^3 = \frac{4}{3}\pi (6.37 \times 10^6 \text{ m})^3 \left( \frac{1 \text{ km}}{10^3 \text{ m}} \right)^3$   
 $= \frac{4}{3}\pi (6.37)^3 \times 10^9 \text{ km}^3 = \underline{\underline{1.08 \times 10^{12} \text{ km}^3}}$

7. Given:  $R = 2000 \text{ km}$

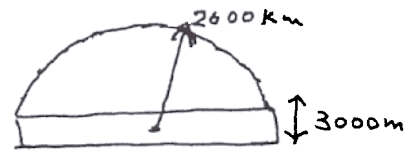
Volume of ice in  $\text{cm}^3$ ?

$$V = \text{Area} \times \text{thickness} = \frac{1}{2}(\pi R^2) \times t$$

$$= \frac{1}{2}\pi (2000 \text{ km})^2 \left( \frac{10^3 \text{ m}}{1 \text{ km}} \right)^2 \left( \frac{10 \text{ cm}}{1 \text{ m}} \right)^2 \times (3000 \text{ m}) \left( \frac{10 \text{ cm}}{1 \text{ m}} \right)$$

$$\frac{1}{2}\pi (2000)^2 \times 10^6 \times 10^4 \text{ cm}^2 \times 3000 \times 10 \text{ cm} = 18.857 \times 10^{21} \text{ cm}^3$$

$$= \underline{\underline{1.9 \times 10^{22} \text{ cm}^3}}$$



11. Conversion:  $1 \text{ m} = 3.281 \text{ ft}$ ,  $1 \text{ s} = 10^9 \text{ ns}$ ;  $1 \text{ m} = 10^3 \text{ mm}$ ,  $1 \text{ s} = 10^{12} \text{ ps}$

$$3 \times 10^8 \text{ m/s} = \left( 3 \times 10^8 \frac{\text{m}}{\text{s}} \right) \left( \frac{3.281 \text{ ft}}{1 \text{ m}} \right) \left( \frac{1 \text{ s}}{10^9 \text{ ns}} \right) = 0.98 \text{ ft/ns}$$

$$3 \times 10^8 \text{ m/s} = \left( 3 \times 10^8 \frac{\text{m}}{\text{s}} \right) \left( \frac{10^3 \text{ mm}}{1 \text{ m}} \right) \left( \frac{1 \text{ s}}{10^{12} \text{ ps}} \right) = 0.30 \text{ mm/ps}$$

Given: Density of Gold =  $19.32 \text{ g/cm}^3$  ( $M = \rho V$ )

a) Volume of 27.63 g of gold =  $\left(\frac{1 \text{ cm}^3}{19.32 \text{ g}}\right) (27.63 \text{ g}) = 1.430 \text{ cm}^3$

$V = \text{Area} \times \text{thickness} = A \times t$

Area =  $\frac{V}{t} = (1.430 \text{ cm}^3) \left(\frac{1}{10^{-6} \text{ m}}\right) \left(\frac{1 \text{ m}}{10^2 \text{ cm}}\right) = \frac{0.0143 \text{ cm}^2}{10^6}$

$A = 14300 \text{ cm}^2 = 1.430 \times 10^4 \text{ cm}^2$

b) Volume of a cylinder =  $\pi R^2 L \Rightarrow L = \frac{V}{\pi R^2}$

$L = (1.430 \text{ cm}^3) \left(\frac{1}{\pi * (2.500 \times 10^{-6} \text{ m})^2 * \left(\frac{10^2 \text{ cm}}{1 \text{ m}}\right)^2}\right) = 0.0728 \text{ cm}$

$L = 7.280 \times 10^{-6} \text{ cm}$

23: Given:  $\rho_{\text{iron}} = 7.87 \text{ g/cm}^3 = (7.87 \frac{\text{g}}{\text{cm}^3}) \left(\frac{1 \text{ kg}}{10^3 \text{ g}}\right) \left(\frac{100 \text{ cm}}{1 \text{ m}}\right)^3$

$M_{\text{iron atom}} = 9.27 \times 10^{-26} \text{ kg}$

a)  $V_{\text{iron atom}} = \frac{M_{\text{iron atom}}}{\rho_{\text{iron}}} = \frac{9.27 \times 10^{-26} \text{ kg}}{7.87 \times 10^3 \text{ kg/m}^3} = 1.18 \times 10^{-29} \text{ m}^3$

The distance between the centers of adjacent atoms is equal to twice the radius, i.e.  $2R$  and  $R$  can be found from the relation



$V_{\text{atom}} = \frac{4}{3} \pi (R_{\text{atom}})^3 \Rightarrow R_{\text{atom}} = \left(\frac{3 V_{\text{atom}}}{4 \pi}\right)^{1/3}$

$= \left(\frac{3 * 1.18 * 10^{-29} \text{ m}^3}{4 \pi}\right)^{1/3} = 0.141 \times 10^{-9} \text{ m}$

$S = 2R = 2.82 \times 10^{-10} \text{ m} = 0.282 \times 10^{-9} \text{ m} = \underline{\underline{0.282 \text{ nm}}}$