

b# 25.

The wavefunction $\Psi(x) = C x e^{-\alpha x^2}$ describes the first excited state of the quantum oscillator.

1) Schrodinger equation in this case is

$$\frac{\partial^2 \Psi(x)}{\partial x^2} = \frac{2m}{\hbar^2} \left(\frac{1}{2} m \omega^2 x^2 - E \right) \Psi(x) \quad \text{--- (1)}$$

$$\begin{aligned} \frac{\partial \Psi}{\partial x} &= C e^{-\alpha x^2} - 2C x^2 \alpha e^{-\alpha x^2} \\ \frac{\partial^2 \Psi}{\partial x^2} &= -2\alpha x C e^{-\alpha x^2} - 4C \alpha x e^{-\alpha x^2} + 4C x^3 \alpha^2 e^{-\alpha x^2} \\ &= -6\alpha (C x e^{-\alpha x^2}) + 4\alpha^2 x^2 (C x e^{-\alpha x^2}) \\ \frac{\partial^2 \Psi}{\partial x^2} &= (-6\alpha + 4\alpha^2 x^2) \Psi(x) \quad \text{--- (2)} \end{aligned}$$

Compare eq. (1) and eq. (2)

$$\Rightarrow \frac{2m}{\hbar^2} E = 6\alpha \quad \text{--- (3)}$$

$$\frac{m^2 \omega^2}{\hbar^2} = 4\alpha^2 \quad \text{--- (4)}$$

$$\Rightarrow \boxed{\alpha = \frac{m\omega}{2\hbar}}$$

$$\text{and } \frac{2mE}{\hbar^2} = 6\alpha = \frac{6m\omega}{2\hbar}$$

$$\Rightarrow \boxed{E = \frac{3}{2} \hbar \omega}$$