

Pb#3.

A free electron has a wavefunction

$$\Psi(x) = A \sin(5 \times 10^{10} x) \quad x \text{ in meters.}$$

a) In general, for a free particle,  $\Psi(x) = A \sin kx$ 

$$\Rightarrow k = 5 \times 10^{10} \text{ m}^{-1}$$

$$\Rightarrow \lambda = \frac{2\pi}{k} = \frac{2\pi}{5 \times 10^{10}} = 1.26 \times 10^{-10} \text{ m} = \boxed{1.26 \text{ \AA}}$$

$$b) \lambda = \frac{h}{p} \Rightarrow p = \frac{h}{\lambda} = \frac{6.63 \times 10^{-34}}{1.26 \times 10^{-10}} = \boxed{5.26 \times 10^{-24} \text{ Kg} \cdot \frac{\text{m}}{\text{s}}}$$

$$c) K = \frac{p^2}{2m} = \frac{(5.26 \times 10^{-24})^2}{2 \times 9.1 \times 10^{-31}} = 1.5 \times 10^{-17} \text{ J} = \boxed{95 \text{ eV}}$$

Pb#6.

The wavefunction of a particle is given by

$$\Psi(x) = A \cos(kx) + B \sin(kx) \quad \text{--- (1)}$$

A, B and k are constants

Schrodinger equation in the case where  $U=0$  is

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x)}{\partial x^2} = E \Psi(x) \quad [\text{time-independent S.E.}]$$

$$\Rightarrow \frac{\partial \Psi}{\partial x} = -A k \sin(kx) + B k \cos(kx)$$

$$\frac{\partial^2 \Psi}{\partial x^2} = -A k^2 \cos(kx) - B k^2 \sin(kx)$$

$$\frac{\partial^2 \Psi}{\partial x^2} = -k^2 \Psi(x) \quad \Rightarrow \quad \frac{2mE}{\hbar^2} = k^2 \quad \text{or} \quad E = \frac{2m}{\hbar^2} k^2$$

So equation (1) is a solution of Schrodinger equation.