

$$n=1000 \quad r_{1000} = 0.0529 \times 10^{-9} \times (1000)^2 = 0.0529 \times 10^{-3} \text{ m}$$

$$f_{1000} = 6.58 \times 10^6 \text{ Hz}$$

$$n=10000 \quad r_{10000} = 0.0529 \times 10^{-9} (10000)^2 = 0.0529 \times 10^{-1} \text{ m}$$

$$f_{10000} = 6.58 \times 10^3 \text{ Hz}$$

$$n_i = n \rightarrow n_f = n-1$$

$$f_n = \frac{\Delta E}{h} = \frac{13.6 \times 1.6 \times 10^{-19}}{6.63 \times 10^{-34}} \left[ \frac{1}{(n-1)^2} - \frac{1}{n^2} \right] \text{ (Hz)}$$

$$= 3.285 \times 10^{15} \left( \frac{1}{(n-1)^2} - \frac{1}{n^2} \right)$$

$$f_{100} = 3.285 \times 10^{15} \left( \frac{1}{99^2} - \frac{1}{100^2} \right) = 6.67 \times 10^9 \text{ Hz}$$

$$f_{1000} = 3.285 \times 10^{15} \left( \frac{1}{999^2} - \frac{1}{1000^2} \right) = 6.58 \times 10^6 \text{ Hz}$$

$$f_{10000} = 3.285 \times 10^{15} \left( \frac{1}{9999^2} - \frac{1}{10000^2} \right) = 6.57 \times 10^3 \text{ Hz}$$

It is clear from this example that when  $n$  becomes large the classical frequency of revolution of the electron calculated in (a) is equal to the quantum frequency calculated in (b)  $\Rightarrow$  the Bohr correspondence principle is valid.