

2. An electron is trapped in an infinitely deep potential well with width $L=0.300$ nm. The wave function for the electron is $\psi(x) = A \sin\left(\frac{n\pi x}{L}\right)$.

- (a) Show that the normalization constant $A = \sqrt{\frac{2}{L}}$
 (b) If the electron is in the ground state, what is the probability of finding the electron between $x = 0.0$ and 0.100 nm.
 (c) Repeat the calculation of part (b) for the $n = 3$ state.
 (d) Calculate the energy in electron-Volt for the electron in the $n = 2$ state.

$$a) \quad 1 = \int_0^L |\psi|^2 dx = A^2 \int_0^L \sin^2 \frac{n\pi x}{L} dx = \frac{A^2}{2} \int_0^L (1 - \cos \frac{2n\pi x}{L}) dx$$

$$= \frac{A^2}{2} \left(x - \frac{L}{2n\pi} \sin \frac{2n\pi x}{L} \right) \Big|_0^L = \frac{A^2 L}{2} \Rightarrow \underline{\underline{A = \sqrt{\frac{2}{L}}}}$$

$$b) \quad \text{Given } L = 0.300 \text{ nm}, n = 1, 0.100 \text{ nm} = \frac{L}{3}$$

$$P_{(0-0.100) \text{ nm}} = \frac{2}{L} \int_0^{L/3} \sin^2 \frac{\pi x}{L} dx = \frac{2}{L} \int_0^{L/3} (1 - \cos \frac{2\pi x}{L}) dx$$

$$P_{0-0.1} = \frac{2}{L} \left(x - \frac{L}{2\pi} \sin \frac{2\pi x}{L} \right) \Big|_0^{L/3} = \frac{2}{L} \left(\frac{L}{3} - \frac{L}{2\pi} \sin \frac{2\pi}{3} - 0 + 0 \right)$$

$$= \frac{2}{3} - \frac{2}{2\pi} \frac{\sqrt{3}}{2} = \underline{\underline{0.1955}}$$

$$c) \quad \text{for } n = 3, \psi = \frac{2}{L} \sin \frac{3\pi x}{L}, 0.100 = \frac{L}{3}$$

$$P_{0-0.1} = \frac{2}{L} \int_0^{L/3} \sin^2 \frac{3\pi x}{L} dx = \frac{2}{L} \int_0^{L/3} (1 - \cos \frac{6\pi x}{L}) dx$$

$$= \frac{2}{L} \left(x - \frac{L}{6\pi} \sin \frac{6\pi x}{L} \right) \Big|_0^{L/3} = \frac{2}{L} \left(\frac{L}{3} - \frac{L}{6\pi} \sin 2\pi - 0 + 0 \right)$$

$$= \frac{2}{3} = \underline{\underline{0.333}}$$

$$d) \quad \text{For } n = 2 \text{ state, we have } 2 \frac{\lambda}{2} = L, \lambda = \frac{h}{p} = \frac{h}{mv}$$

$$K = \frac{p^2}{2m} = \frac{h^2}{2mL^2} = \frac{(6.62 \times 10^{-34} \text{ J}\cdot\text{s})^2}{2 \times 9.11 \times 10^{-31} \text{ kg} (0.300 \times 10^{-9} \text{ m})^2}$$

$$= 2.68 \times 10^{-18} \text{ J} = \underline{\underline{16.7 \text{ eV}}}$$