

Exercise 2 :

$$E_1 = \frac{1}{2} m_e v^2 = \frac{\pi^2 \hbar^2}{2m_e L^2} \Rightarrow v = \sqrt{\frac{\pi^2 \hbar^2}{m_e L^2}}$$

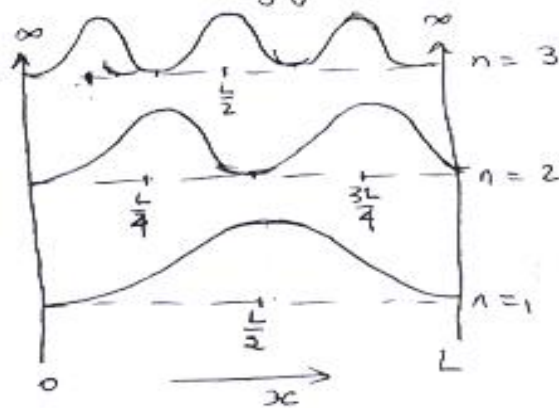
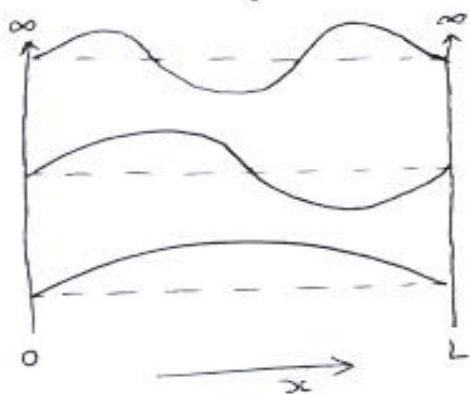
$$\hbar = \frac{h}{2\pi} \quad v = \sqrt{\frac{h}{4m_e L^2}} =$$

Return to the wavefunction

$$k = \frac{n\pi}{L} \quad \text{and} \quad B=0 \Rightarrow \boxed{\Psi_n(x) = A \sin\left(\frac{n\pi x}{L}\right)} \quad 0 \leq x \leq L$$

$n=1, 2, \dots$

For each value of the quantum number  $n$  there is a wavefunction  $\Psi_n(x)$  describing the state of the particle with energy  $E_n$ .



for  $n=1$   $P$  is largest at  $x = \frac{L}{2}$  - this is the most probable position.

for  $n=2$   $P$  is maximum at  $x = \frac{L}{4}$  and  $\frac{3L}{4}$

for  $n=3$   $P$  is ~~maximum~~ <sup>minimum</sup> at  $x = \frac{L}{3}, \frac{2L}{3}, \dots$  It is impossible to find the particle at those locations.

Let us normalize the wavefunction  $\Psi_n$

$$1 = \int_{-\infty}^{+\infty} |\Psi_n(x)|^2 dx = A^2 \int_0^L \sin^2\left(\frac{n\pi x}{L}\right) dx$$

$$1 = \frac{A^2}{2} \int_0^L (1 - \cos\left(\frac{2\pi n x}{L}\right)) dx = A^2 \frac{L}{2} \Rightarrow A = \underline{\underline{\sqrt{\frac{2}{L}}}}$$