

If  $v = 3 \text{ cm/s}$   $E = \frac{1}{2} m v^2 = 4.5 \times 10^{-10} \text{ J}$

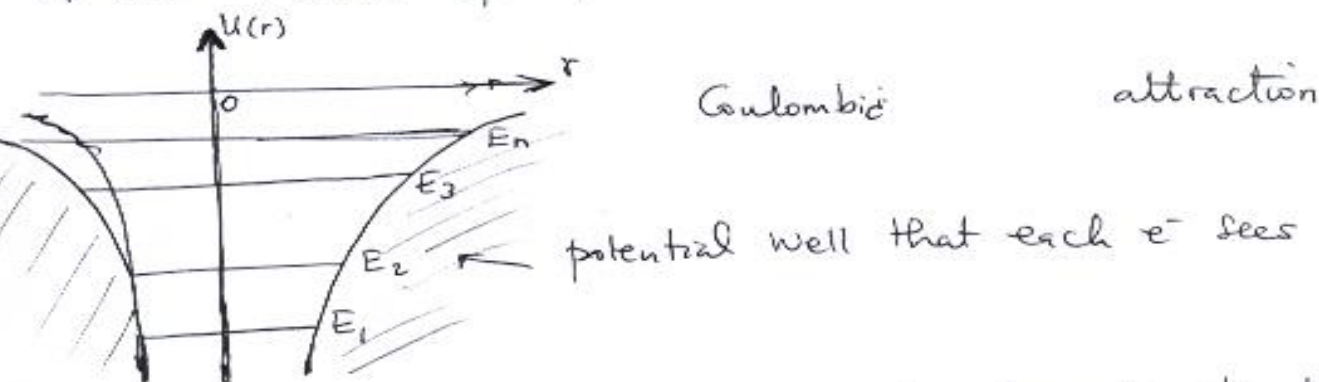
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$n = \sqrt{\frac{8mL^2E}{h}} = 9.05 \times 10^{23}$   $n$  is very large

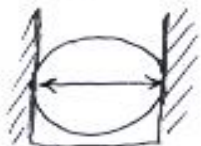
We are in the classical

region.

Ex 5.7 Model of an Atom



Use the model of a particle in a box to estimate the energy (in eV) required to raise an atomic  $e^-$  from state  $n=1$  to state  $n=2$  assume the atom's radius =  $0.1 \text{ nm} = 1 \text{ \AA}$



$L = 0.2 \text{ nm}$

$m_e = 0.511 \frac{\text{MeV}}{c^2}$

$hc = 0.1973 \text{ keV}$

$E_1 = \frac{\pi^2 \hbar^2}{2m_e L^2} = 9.40 \text{ eV}$

$E_2 = 4 \cdot E_1 = 37.6 \text{ eV}$

$\Delta E = E_2 - E_1 = 28.2 \text{ eV}$

Calculate  $\lambda$  !  $\Delta E = hf = h \frac{c}{\lambda} \Rightarrow \lambda = \frac{hc}{\Delta E} = \frac{12400}{28.2} = 440 \text{ \AA}$

in the far UV region.

This is just a model that is valid for low  $n$  but for high  $n$  it gets worse!