

$$\Psi(0) = 0 \quad x = L \quad \Rightarrow \quad A \sin kL = 0$$

$$\Rightarrow \quad kL = n\pi \quad n = 1, 2, 3, \dots$$

$$\left[k = \frac{2\pi}{\lambda} \Rightarrow \frac{\lambda}{2\pi} L = n\pi \Rightarrow \frac{\lambda}{2} = \frac{n\pi^2}{L} \right]$$

$$k = \frac{n\pi}{L} = \frac{\sqrt{2mE}}{\hbar} \Rightarrow \boxed{E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}} \quad n = 1, 2, 3, \dots$$

n is the quantum number called

energy is quantized

$n=1 \rightarrow$ The ground state corresponds to $n=1$

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} \quad \text{or zero-point energy} > 0$$

E_1 and $E_n = n^2 E_1$, $n = 2, 3, 4, \dots$ are the excited states. [Do not think that the particle jump from one \bar{t} to another]

$E=0$ is not allowed; the particle can never be at rest.

$$n=0 \quad E=0 \quad \frac{d^2\Psi}{dx^2} = 0 \Rightarrow \Psi = ax + b$$

$$\Psi(0) = 0 \Rightarrow b = 0 \quad \Psi(L) = 0 \Rightarrow a = 0 \Rightarrow \Psi = 0$$

everywhere. $|\Psi|^2 = 0$ the particle cannot be found \Rightarrow

no description is possible when $E=0$. Classically $E=0$ is allowed and E is continuous.

Ex 5.6 Energy Quantization for Macroscopic Objects.

$$m = 1.00 \text{ mg} \quad L = 1.00 \text{ cm}$$

$$a) \quad E_1 = \frac{\pi^2 \hbar^2}{2mL^2} = \frac{1}{2} m v^2 \Rightarrow v = \sqrt{\frac{2\pi^2 \hbar^2}{2m^2 L^2}} = \sqrt{\frac{\hbar^2}{4mL^2}}$$

$$\hbar = \frac{h}{2\pi} \quad v = \sqrt{\frac{(6.63 \times 10^{-34})^2}{4(1 \times 10^{-3})(0.01)^2}} = 3.31 \times 10^{-26} \text{ m/s}$$

This is too small. The object can be considered to be at rest.