

Chapter 2: The Quantum Theory of Light

2.2 Assume that your skin can be considered as a blackbody. We can use Wien's displacement law:

$$\lambda_{\max} T = 2.898 \times 10^{-3} \text{ m} \cdot \text{K}$$

$$\text{When } T = 35^\circ\text{C} = 308 \text{ K} \Rightarrow \lambda_{\max} = \frac{2.898 \times 10^{-3}}{308}$$

$$\Rightarrow \lambda_{\max} = 9.41 \times 10^{-6} \text{ m} = \boxed{9410 \text{ nm}}$$

2.9

$$P = \frac{E}{t} = n \frac{hf}{t} \Rightarrow n = \frac{P t}{hf}$$

$$t = 1 \text{ sec} \Rightarrow n = \frac{100 \times 10^3}{6.62 \times 10^{-34} \times 94 \times 10^6} = \boxed{1.6 \times 10^{30} \text{ photons/sec}}$$

2.14

In general $K = hf - \phi$

a) at the cut-off wavelength $K = 0 \Rightarrow \frac{hc}{\lambda_c} = \phi$

$$\Rightarrow \lambda_c = \frac{hc}{\phi} = \frac{1240 \text{ eV} \cdot \text{nm}}{4.2 \text{ eV}} = \boxed{295 \text{ nm}}$$

The threshold frequency is given by

$$f_0 = \frac{c}{\lambda_c} = \frac{3 \times 10^8}{295 \times 10^{-9}} = \boxed{1 \times 10^{15} \text{ Hz}}$$

b)

$$K = eV_s = hf - \phi$$

$$\begin{aligned} \Rightarrow V_s &= \frac{hf - \phi}{e} = \frac{hc}{\lambda e} - \frac{\phi}{e} \\ &= \frac{1240 \text{ eV nm}}{200 \text{ nm} \times 1e} - \frac{4.2 \text{ eV}}{1e} \end{aligned}$$

$$\boxed{V_s = 2.0 \text{ V}}$$

2.24

$$E = 300 \times 10^3 \text{ eV} \quad \theta = 30^\circ \text{ (scattering angle)}$$

$$\begin{aligned} \text{a) } \Delta \lambda &= \lambda' - \lambda_0 = \frac{h}{m_e c} (1 - \cos \theta) \\ &= (0.00243 \text{ nm}) (1 - \cos 30^\circ) = \boxed{3.26 \times 10^{-4} \text{ nm}} \end{aligned}$$

b)

$$E' = \frac{hc}{\lambda'}$$

$$\begin{aligned} \lambda' &= \lambda_0 + \Delta \lambda = \frac{hc}{E} + \Delta \lambda = \frac{1240 \text{ eV nm}}{3 \times 10^5 \text{ eV}} + 3.26 \times 10^{-4} \text{ nm} \\ &= 4.46 \times 10^{-3} \text{ nm} \end{aligned}$$

$$\Rightarrow E' = \frac{1240 \text{ eV nm}}{4.46 \times 10^{-3} \text{ nm}} = \boxed{2.78 \times 10^5 \text{ eV}}$$

c)

$$\begin{aligned} E &= K_e + E' \Rightarrow K_e = E - E' \\ &= 3 \times 10^5 - 2.78 \times 10^5 \text{ eV} \\ &= \boxed{22 \times 10^3 \text{ eV}} \end{aligned}$$