

**First major exam Phys -102 Term 992**

(1) Q0 A transverse sinusoidal wave travels along a string of linear mass density 5.00 g/m. The amplitude of the wave is 2.00 cm, its frequency is 60.0 Hz, and the tension in the string is 20.0 N. What is the power transmitted by this wave?

Q0

A1 8.99 W

A2 90.0 W

A3 285 W

A4 2.54 W

A5 512 W

Q0

(2) Q0 The equation of a wave traveling along a string, under a tension of 10 N, is given by:

Q0  $y = (6.0 \text{ cm}) \sin(0.02\pi x + 40.0\pi t)$ ,

Q0 where  $x$  is in centimeters and  $t$  is in seconds.

Q0 Determine the mass per unit length of the string.

Q0

A1 25 g/m

A2 60 g/m

A3 50 g/m

A4 10 g/m

A5 72 g/m

Q0

(3) Q0 A transverse sinusoidal wave traveling in the negative  $x$  direction has an amplitude of 10.0 cm, a wavelength of 20.0 cm, and a frequency of 8.00 Hz. Write the expression for  $y$  as a function of  $x$  (in meters) and  $t$  (in seconds) if  $y(0,0) = 10.0 \text{ cm}$ .

Q0

A1  $y = (0.1 \text{ m}) \sin[31.4x + 50.3t + (\pi/2)]$

A2  $y = (0.1 \text{ m}) \sin[20.0x - 8.00t - (2\pi)]$

A3  $y = (0.1 \text{ m}) \sin[20.0x + 8.00t + (2\pi)]$

A4  $y = (0.1 \text{ m}) \sin[31.4x - 50.3t - (\pi/2)]$

A5  $y = (0.1 \text{ m}) \sin[31.4x + 50.3t + \pi]$

Q0

(4) Q0 Transverse waves, with fixed amplitude, are being generated on a rope under constant tension. When the frequency of the wave is increased, which one of the following statements

Q0 is correct:

Q0

A1 The wavelength decreases and the transmitted power increases

A2 The wavelength increases and the transmitted power is the same

A3 The maximum transverse speed is the same and the transmitted

A3 power increases

A4 Both the wavelength and the linear mass density decrease

A5 Both the wavelength and the maximum transverse speed increase  
Q0

(5)Q0 An ambulance siren emits a sound of frequency 1.60 kHz. A  
17 Q0 person running with a speed of 2.50 m/s hears a frequency of  
Q0 1.70 kHz as the ambulance approaches him from the back. How  
Q0 fast is the ambulance moving? (speed of sound is 340 m/s).

Q0

A1 22.4 m/s

A2 17.7 m/s

A3 12.2 m/s

A4 25.6 m/s

A5 2.50 m/s

Q0

(6)Q0 A source of sound (1000 Hz) emits uniformly in all directions.  
17 Q0 An observer 3.0 m from the source measures a sound level of  
Q0 40 dB. Calculate the average power output of the source.

Q0

A1 1.13 micro-W

A2 2.87 micro-W

A3 0.34 micro-W

A4 10.5 micro-W

A5 5.23 micro-W

Q0

(7)Q0 Consider two sound waves A and B propagating in the same  
17 Q0 medium. Find the ratio of the intensity of the sound wave A  
Q0 to the intensity of the sound wave B if the sound level of  
Q0 wave A is 20 dB greater than the sound level of wave B.

Q0

A1 100

A2 20

A3 5

A4 15

A5 10

Q0

(8)Q0 The equation for a standing wave is given by:

18 Q0  $y = 4.00 \cdot 10^{**}(-3) \sin(2.09 x) \cos(60.0 t)$  (SI units).

Q0 What is the distance between two consecutive antinodes?

Q0

A1 1.50 m

A2 0.560 m

A3 2.20 m

A4 5.00 m

A5 3.00 m

Q0

(9)Q0 The frequencies of two consecutive harmonics of a pipe, closed  
18 Q0 at one end, are 929 Hz and 1300 Hz. The length of the pipe is

Q0 0.450 m. What is the speed of the wave in the pipe?

Q0

A1 334 m/s

A2 343 m/s

A3 350 m/s

A4 340 m/s

A5 348 m/s

Q0

10 Q0 Two speakers are driven by a common oscillator and face each

18 Q0 other at a distance of 1.500 m. A man is standing at 0.700 m

Q0 from one of the speakers along the line joining the two

Q0 speakers. What is the highest frequency of the oscillator,

Q0 within the audible range (20.0 Hz to 20.0 kHz), so that the

Q0 man hears a minimum sound? (Speed of sound = 343 m/s).

Q0

A1 18.9 kHz

A2 19.9 kHz

A3 10.3 kHz

A4 12.6 kHz

A5 15.9 kHz

Q0

11 Q0 Fully DESTRUCTIVE interference between two sinusoidal waves

18 Q0 of the same frequency and amplitude occurs only if they:

Q0

A1 travel in the same direction and are 180 degrees out of phase

A2 travel in opposite directions and are 90 degrees out of phase

A3 travel in opposite directions and are in phase

A4 travel in the same direction and are 90 degrees out of phase

A5 travel in the same direction and are in phase

Q0

12 Q0 Which one of the following statements is wrong?

19 Q0

A1 Two bodies can be in thermal contact for a very long time

A1 without being in thermal equilibrium

A2 The density of most substances decreases when they are heated

A3 If two bodies are in thermal equilibrium than they must have

A3 the same temperature

A4 Generally liquids expand more than solids for the same

A4 temperature change

A5 Most solid materials contract when cooled

Q0

13 Q0 The volume expansion coefficient of benzene is

19 Q0  $1.24 \times 10^{-4} / \text{C-degree}$ . If a  $100 \text{ cm}^3$  steel container is

Q0 filled with benzene when the temperature is 20.0 degrees-C,

Q0 how much benzene will spill over when the temperature is

Q0 raised to 50.0 degrees-C?

Q0 (neglect the thermal expansion of the steel container.)

Q0

A1  $0.372 \text{ cm}^3$

A2  $0.436 \text{ cm}^3$

A3  $0.124 \text{ cm}^3$

A4  $0.865 \text{ cm}^3$

A5  $0.236 \text{ cm}^3$

Q0

14 Q0 A steel vessel contains 5 moles of an ideal gas at 0 degree-C

19 Q0 and a pressure of 1 atm. It is heated at constant volume until

Q0 its temperature is 100 degrees-C. How many moles of gas should

Q0 be removed from the container to keep the pressure of the gas

Q0 constant at 1 atm?

Q0

A1 1.34 moles

A2 0 moles

A3 4.32 moles

A4 3.66 moles

A5 2.45 moles

Q0

15 Q0 How much ice at -20 degrees-C must be mixed with 0.25 kg of

20 Q0 water, initially at 20 degrees-C, in order for the final

Q0 temperature to be 0 degrees-C with the ice all melted?

Q0 (neglect the heat capacity of the container.)

Q0  $C(\text{water}) = 4186 \text{ J/Kg C-degrees}$ ,  $C(\text{ice}) = 2000 \text{ J/kg C-degrees}$

Q0  $L_f(\text{ice}) = 33.4 \times 10^4 \text{ J/kg}$ .

Q0

A1 56 g

A2 63 g

A3 85 g

A4 75 g

A5 12 g

Q0

16 Q0 Which one of the following statements is true?

20 Q0

A1 In a cyclic process, the change in internal energy is zero

A2 In an adiabatic process, the heat flow is positive

A3 In an isovolumetric process, the work done is positive

A4 The internal energy of a system is not a state function

A5 In an isobaric process, the change in internal energy is

A5 always zero

Q0

17 Q0 Consider an isothermal compression of 0.1 moles of an ideal

20 Q0 gas at a temperature of 0 degree-C. The initial pressure of

Q0 the gas is 1 atm and the final volume is 1/5 the initial

Q0 volume. Find the thermal energy transfer for this process.

Q0

A1 365 J lost by the gas

A2 365 J gained by the gas

A3 0 J

A4 124 J lost by the gas

A5 124 J gained by the gas

Q0

18 Q0 A glass window has an area of  $0.50 \text{ m}^2$  and a thickness

20 Q0 of  $0.60 \text{ cm}$ . If the rate of heat flow between the faces is

Q0  $500 \text{ kJ/hour}$ , find the temperature difference between the

Q0 window's faces.  $K(\text{glass}) = 0.80 \text{ W/m C-degrees}$ .

Q0

A1  $2.1 \text{ C-degrees}$

A2  $3.5 \text{ C-degrees}$

A3  $1.2 \text{ C-degrees}$

A4  $12 \text{ C-degrees}$

A5  $45 \text{ C-degrees}$

Q0

19 Q0 A cylinder contains 4 moles of a diatomic ideal gas ( $C_v = 5R/2$ )

21 Q0 at a temperature of  $27 \text{ degrees-C}$  and a pressure of  $1.5 \text{ atm}$ .

Q0 The gas is heated under constant pressure until its

Q0 temperature reaches  $127 \text{ degrees-C}$ . How much work is done by

Q0 the gas in this process?

Q0

A1 794 calories

A2 418 calories

A3 150 calories

A4 562 calories

A5 986 calories

Q0

20 Q0 Consider  $100 \text{ g}$  of helium (He) gas at  $77 \text{ K}$ . How much heat

21 Q0 energy must be supplied to the gas to increase its

Q0 temperature to  $24 \text{ degrees-C}$ , if the process is isovolumetric?

Q0 ( $M(\text{He}) = 4 \text{ g/mole}$  and He is a monatomic gas.)

Q0

A1  $69 \text{ kJ}$

A2  $71 \text{ kJ}$

A3  $43 \text{ kJ}$

A4  $24 \text{ kJ}$

A5  $12 \text{ kJ}$