

Q1.

If you setup the fifth harmonic on a string clamped at both ends, is there a node, antinode or some intermediate state at the midpoint of the string?

- A) Antinode
- B) Node
- C) Intermediate state
- D) None of the others
- E) Not enough information given

Q2.

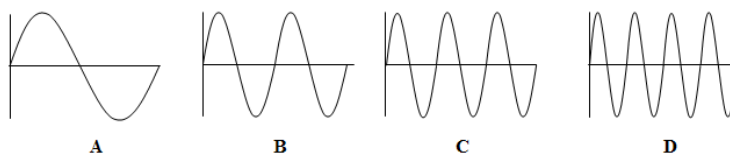
A string of 80 cm length is fixed at both ends. The string oscillates in the fundamental mode with a frequency of 60 Hz and a maximum amplitude of 0.3 cm of the standing wave. What is the maximum transverse speed of a particle oscillating on the string at $x=20$ cm?

- A) 80 cm/s
- B) 71 cm/s
- C) 66 cm/s
- D) 99 cm/s
- E) 91 cm/s

Q3.

A string fixed at both ends can be made to vibrate in one of the four patterns shown in **FIGURE 1** by varying the tension in the string. Arrange the four patterns in terms of the average power transported by the wave in the string, **smallest first**. The frequency and amplitude of the waves are the same in all the figures.

Fig#



- A) D, C, B, A
- B) A, C, B, D
- C) B, C, D, A
- D) C, B, A, D
- E) D, C, A, B

Q4.

The displacement of a string carrying a traveling sinusoidal wave is given by

$$y(x,t) = y_m \sin(kx - \omega t + \phi), \text{ where } x \text{ is in meters and } t \text{ is in seconds.}$$

At time $t = 0$ a particle at $x = 0$ has transverse speed u_0 and displacement y_0 . Then magnitude of $\tan\phi$ is equal to:

- A) $\omega y_0 / u_0$
- B) $u_0 / \omega y_0$
- C) $\omega u_0 / y_0$
- D) $y_0 / \omega u_0$
- E) $\omega u_0 y_0$

Q5.

A sound meter placed 3.0 m from a point sound source registers a sound level of 80 dB. What sound level will the sound meter register if the power of the source is reduced by a factor of 25?

- A) 66 dB
- B) 11 dB
- C) 32 dB
- D) 3.2 dB
- E) 55 dB

Q6.

A pipe of length L , closed at one end, is resonating at its fundamental frequency. Which one of the following statements is **TRUE**?

- A) The wavelength is $4L$ and there is a displacement antinode at the pipe's open end
 - B) The wavelength is $4L$ and there is a displacement node at the pipe's open end
 - C) The wavelength is $2L$ and there is a displacement node at the pipe's open end
 - D) The wavelength is $2L$ and there is a displacement antinode at the pipe's open end
 - E) The wavelength is L and there is a displacement antinode at the pipe's open end
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Q7.

Two loudspeakers S_1 and S_2 are in phase and emit sound waves with the same frequency. They are placed along the y -axis and are separated by a distance of 8.00 m, as shown in **FIGURE 2**. A person is standing at point O which is 12.0 m from the y -axis and equidistant from the loudspeakers. When the person moves from point O to point P at a distance of 3 m, he detects the second destructive interference in sound intensity. What is the frequency of the sound waves emitted by the loudspeakers? The speed of sound in air is 343 m/s.

Fig#

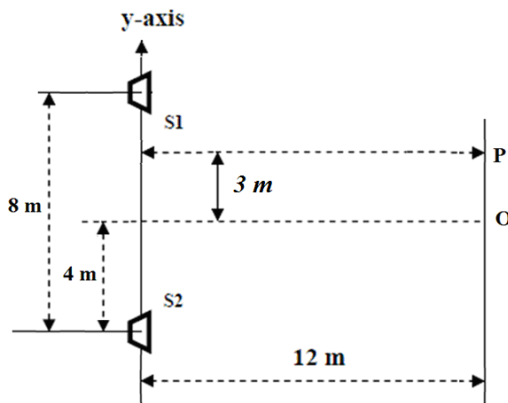


FIGURE NOT TO SCALE

- A) 278 Hz
- B) 178 Hz
- C) 222 Hz
- D) 335 Hz
- E) 522 Hz

Q8.

A bat emits sound at a frequency of 3.00×10^4 Hz as it approaches a wall. The frequency of the sound reflected from the wall and detected by the bat is 3.09×10^4 Hz. What is the speed of the bat? The speed of sound in air is 343 m/s.

- A) 5.07 m/s
- B) 3.50 m/s
- C) 2.20 m/s
- D) 6.30 m/s
- E) 7.70 m/s

Q9.

Materials A, B, and C are solids that are at their melting temperatures. Material A requires 200 J to melt 4 kg, material B requires 300 J to melt 5 kg, and material C requires 300 J to melt 6 kg. Rank the materials according to their heats of fusion, **greatest first**.

- A) B, then A and C tie
- B) A, then B and C tie
- C) C, then A and B tie
- D) A, B and C all tie
- E) None of the others

Q10.

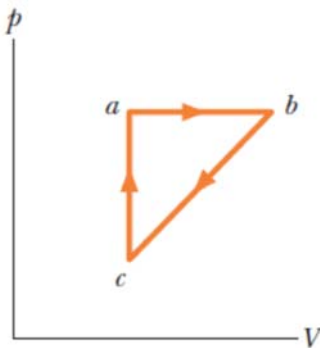
What is the volume of a lead ball at 10.0 °C if the ball's volume at 260 °C is 97.0 cm³? (Coefficient of linear expansion of lead $\alpha_{Pb} = 29.0 \times 10^{-6} / ^\circ\text{C}$)?

- A) 94.9 cm³
- B) 96.1 cm³
- C) 95.5 cm³
- D) 92.1 cm³
- E) 93.5 cm³

Q11.

The net work done by a gas, when taken through cycle **abca**, as shown in the p-V diagram of **FIGURE 3** is +2.1 J. Along path **ab**, the change in the internal energy is +3.2 J and the magnitude of the work done is 5.9 J. Along path **ca**, the energy transferred to the gas as heat is +1.6 J. What is change in the internal energy and how much energy is transferred as heat along path **bc**?

Fig#

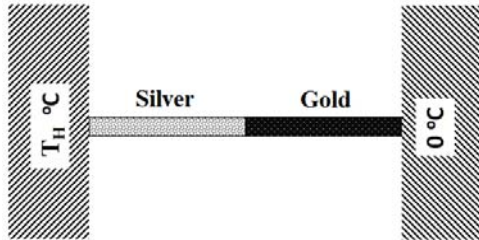


- A) -4.8 J and -8.6 J
- B) -2.5 J and -5.1 J
- C) +8.0 J and +7.5 J
- D) +5.5 J and +4.3 J
- E) -8.0 J and -9.0 J

Q12.

Two metal cylindrical rods, one gold and the other silver, are welded end-to-end and placed between two heat reservoirs at 0.00°C and T_H $^\circ\text{C}$ temperatures, as shown in **FIGURE 4**. Each rod is 5.00 cm long and has a cross sectional area of 4.00 cm^2 . What is the temperature T_H of the hot reservoir if, in the steady state, 8.20 kJ of heat flows through the two rods in 60.0 seconds? (Thermal conductivities: $k_{\text{Silver}} = 417\text{ W/m.K}$, $k_{\text{Gold}} = 219\text{ W/m.K}$)

Fig#



- A) 119°C
- B) 103°C
- C) 100°C
- D) 127°C
- E) 135°C

Q13.

A certain amount of an ideal gas absorbs 30 J of heat at constant volume when its temperature increases by ΔT $^\circ\text{C}$. When the same gas is heated at constant pressure it absorbs 50 J of heat for the same ΔT $^\circ\text{C}$. How much work is done by the gas in the constant pressure process?

- A) 20 J
- B) 33 J
- C) 50 J
- D) 15 J
- E) 10 J

Q14.

What is the percentage decrease in v_{rms} of the molecules of an ideal hydrogen gas if its temperature is reduced from 100°C to 20°C ?

- A) 11%
- B) 7%
- C) 3%
- D) 15%
- E) 14%

Q15.

If W is the magnitude of work done on an ideal diatomic gas in an adiabatic process, then the change in translational kinetic energy of the gas molecules is:

- A) $3W/5$ J
- B) $2W/5$ J
- C) $5W/2$ J
- D) 0 J
- E) W J

Q16.

An ideal diatomic gas occupies a volume of 3.50 L at a pressure of 1.20 atm and a temperature of 300 K. It is compressed adiabatically to a volume of 0.55 L. What is the magnitude of work done in this adiabatic process?

- A) 1.16 kJ
- B) 0.25 kJ
- C) 2.10 kJ
- D) 3.61 kJ
- E) 1.00 kJ

Q17.

What will happen to the entropy of an ideal gas that expands in an isothermal process?

- A) It will increase.
- B) It will decrease.
- C) It will remain unchanged.
- D) Need more information to answer.
- E) Entropy change is not defined for an isothermal process.

Q18.

The change in entropy of 20.0 moles of an ideal monatomic gas in a constant volume process is 200 J/K. If the initial temperature of the gas was 300 K, what is its final temperature?

- A) 669 K
- B) 562 K
- C) 427 K
- D) 187 K
- E) 345 K

Q19.

A Carnot engine whose hot reservoir temperature is 400°C has a thermal efficiency of 40 %. By how many degrees should we lower the temperature of the cold reservoir to increase the engine efficiency to 60%?

- A) 135°C
- B) 105°C
- C) 215°C
- D) 119°C
- E) 171°C

Q20.

The operating temperature of a Carnot refrigerator is 2.0°C . The refrigerator is placed in a kitchen where the temperature is 22°C . What power is needed to operate this refrigerator in order to extract from it 89 MJ of heat in one hour?

- A) 1.8 kW
 - B) 2.0 kW
 - C) 1.5 kW
 - D) 2.9 kW
 - E) 1.0 kW
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