

**Vibrational spectra and potential energy distributions of normal modes of 3-nitroso- and 3-nitrocyclopropenes.** Badawi, Hassan M.; El-Rayyes, Ali A. Department of Chemistry, King Fahd University of Petroleum and Minerals, Dhahran, Saudi Arabia. THEOCHEM (2002), 588 17-27.

### **Abstract**

The structural stability of 3-nitrosocyclopropene  $c\text{-C}_3\text{H}_3\text{-N:O}$  and 3-nitrocyclopropene  $c\text{-C}_3\text{H}_3\text{-NO}_2$  was studied by density functional B3LYP and ab initio MP2 calculations using 6-311+G\*\* basis set. From the calculations, 3-nitrosocyclopropene was predicted to exist predominantly in the trans conformation (N:O bond is trans to the ring and eclipses  $\alpha\text{-H}$  of the ring) with high trans-cis rotational barrier of approximately 8 kcal/mol. The NO<sub>2</sub> rotational barrier in 3-nitrocyclopropene was predicted from the symmetric potential scan to be of approximately 4.4 kcal/mol. The vibrational frequencies were computed at DFT-B3LYP/6-311+G\*\* level for both molecules. Normal coordinate calculations were carried out and potential energy distributions were calculated for the two molecules.