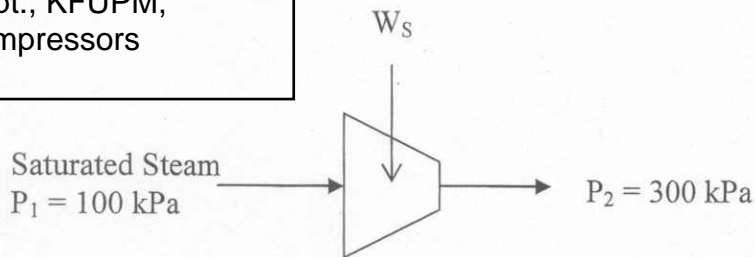


Example 7.8

Saturated steam at 100 kPa is compressed adiabatically to 300 kPa. If the compressor efficiency is 0.75, what is the work required?

Chemical Engineering Dept., KFUPM,
CHE303, Handout_7, Compressors



$$S_1 = 7.3598 \frac{\text{kJ}}{\text{kg K}}$$

$$H_1 = 2675.4 \frac{\text{kJ}}{\text{kg}}$$

Step #1 Assume isentropic compressor ($\Delta S = 0$)

$$\Rightarrow S_2' = S_1 = 7.3598 \text{ kJ/(kg K)}$$

$$\Rightarrow P_2 = 300 \text{ kPa} \quad S_2' = 7.3598$$

note at 300 kPa $S_2^v = 6.9909 < S_2'$

\Rightarrow steam at discharge is superheated

$$\text{now, } P_2 = 300 \text{ kPa} \quad S_2' = 7.3598 \text{ kJ/(kg K)}$$

$$\Rightarrow T_2' = 211.3^\circ \text{C} \quad (\text{by interpolation})$$

$$\text{and } H_2' = 2888.8 \frac{\text{kJ}}{\text{kg}} \quad (\quad \text{ " } \quad \text{ " } \quad)$$

$$(\Delta H)_s = H_2' - H_1 = 213.4 \frac{\text{kJ}}{\text{kg}}$$

$$H_2 = \frac{(\Delta H)_s}{2} + H_1$$

$$= \frac{213.4}{0.75} + 2675.4 = 2959.9 \frac{\text{kJ}}{\text{kg}}$$

$$\Rightarrow \left. \begin{array}{l} P_2 = 300 \text{ kPa} \\ H_2 = 2959.9 \frac{\text{kJ}}{\text{kg}} \end{array} \right\} \Rightarrow \begin{array}{l} T_2 = 246.1 \text{ } ^\circ\text{C} \\ S_2 = 7.5019 \frac{\text{kJ}}{\text{kg K}} \end{array}$$

$$W_s = H_2 - H_1 = 284.5 \frac{\text{kJ}}{\text{kg}}$$